



# Introduction to Quantum Materials

Leon Balents, KITP

QS3 School, June 11, 2018

# Quantum Materials

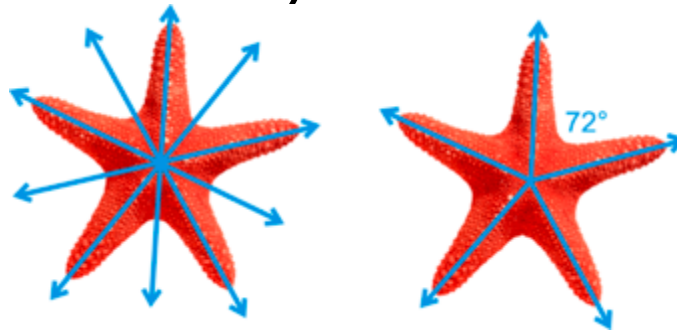
- What are they? Materials where electrons are doing interesting quantum things
- The plan:
  - Lecture 1: Concepts in Quantum Materials
  - Lecture 2: Survey of actual materials

# Themes of modern QMs

- Order
- Topology
- Entanglement
- Correlations
- Dynamics

# Order: symmetry

- Symmetry: a way to organize matter
- A symmetry is some operation that leaves a system (i.e. a material) invariant (unchanged



- In physics, we usually mean it leaves the Hamiltonian invariant

$$U^\dagger H U = H$$

# Order and symmetry

- Why symmetry?
  - It is *persistent*: it only changes through a phase transition
  - It has numerous implications:
    - Quantum numbers and degeneracies
    - Conservation laws
  - Brings powerful mathematics of group theory
    - The set of all symmetries of a system form its symmetry group. Materials with different symmetry groups are in different *phases*

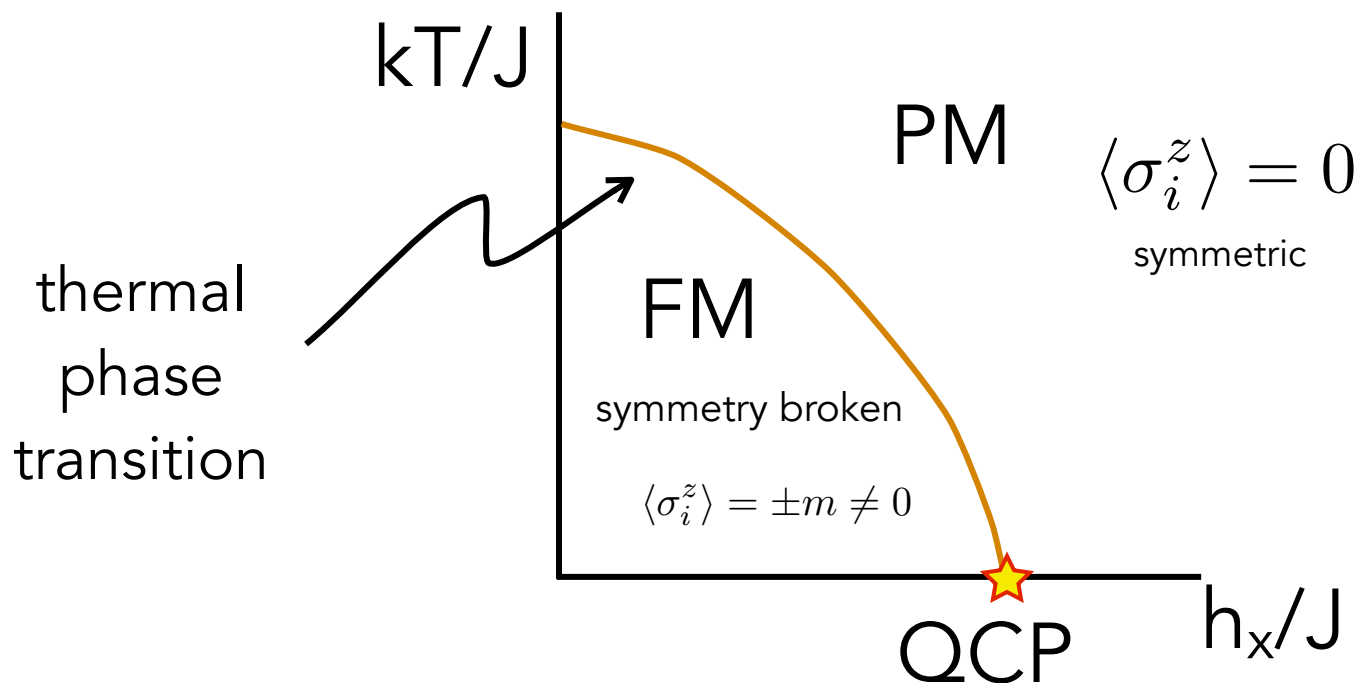
# Ising model

- A canonical example

$$H = -J \sum_{\langle ij \rangle} \sigma_i^z \sigma_j^z - h_x \sum_i \sigma_i^x$$

$$\sigma^z \rightarrow -\sigma^z$$

$Z_2$  symmetry



$\psi = \langle \sigma_i^z \rangle$

"order  
parameter"

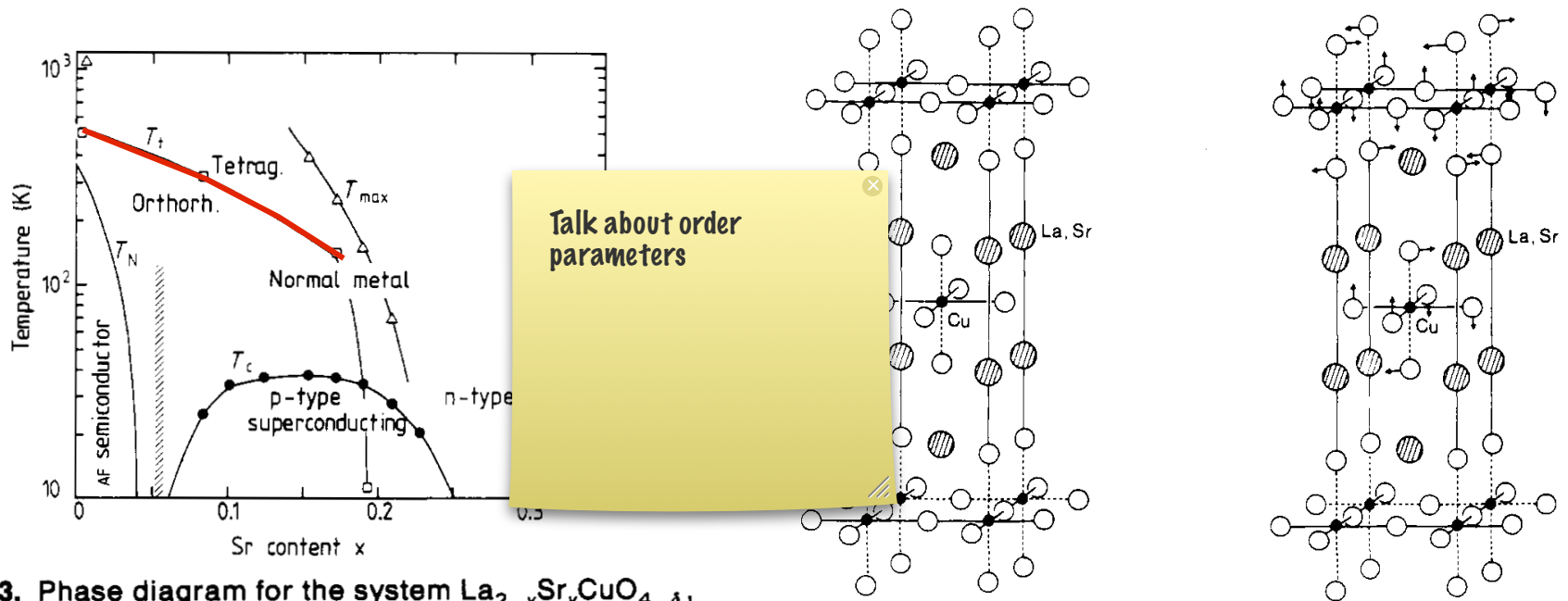
# Symmetries in QMs

- Basic symmetries of our world:
  - space-time (Lorentz/Poincare) symmetry
    - spatial isotropy and translations
    - time reversal
  - Charge/particle number conservation
- Approximate symmetries (sometimes)
  - spin-rotation
  - various internal quantum numbers
- These things are broken down to varying degrees in different QMs

# Crystallography

- Crystal structure: 230 crystallographic space groups
  - Classifies the arrangements of atoms (which break the symmetries of free space)
  - Wallpaper groups in 2d (c.f. 2d materials)
  - Basic input to many things
    - Phonons, elasticity, band structure...
  - LOADS of extremely useful stuff on Bilbao crystallographic server...
- Structural phase transition = change of space group.

# Structural phase transition



**Figure 3.** Phase diagram for the system  $\text{La}_{2-y}\text{Sr}_y\text{CuO}_{4-\delta}$ ,  $p = y - 2\delta$ . Adapted from [15].

tetragonal    orthorhombic

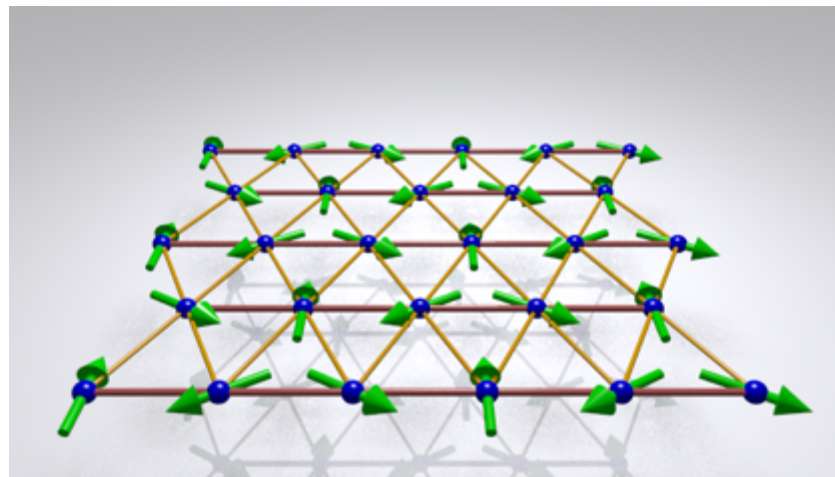
"Order parameters" ~ soft  
phonon modes

# Magnetism

- Fundamental symmetry is *time-reversal*

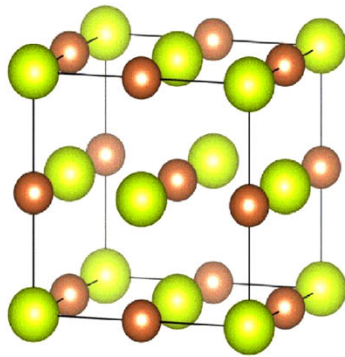
$$\mathbf{S}_i \rightarrow -\mathbf{S}_i$$

- Any ordering of magnetic moments breaks this symmetry

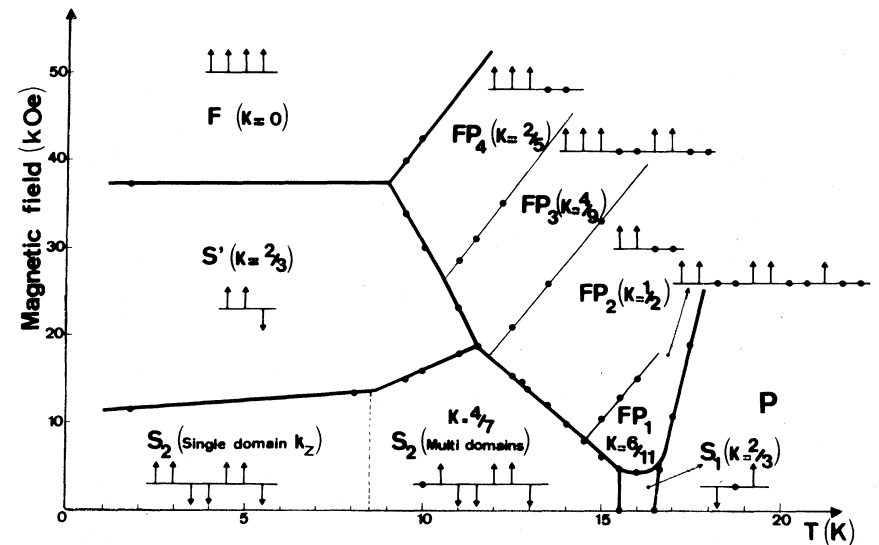
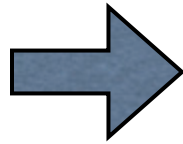


# Magnetism

- Broken T-reversal systems are very rich:  
1651 magnetic space groups



CeSb



Rossat-Mignod et al, 1977

# Superconductivity

- Charge conservation symmetry is spontaneously broken in a SC

- Order parameter  $\sim$  "pair wavefunction"

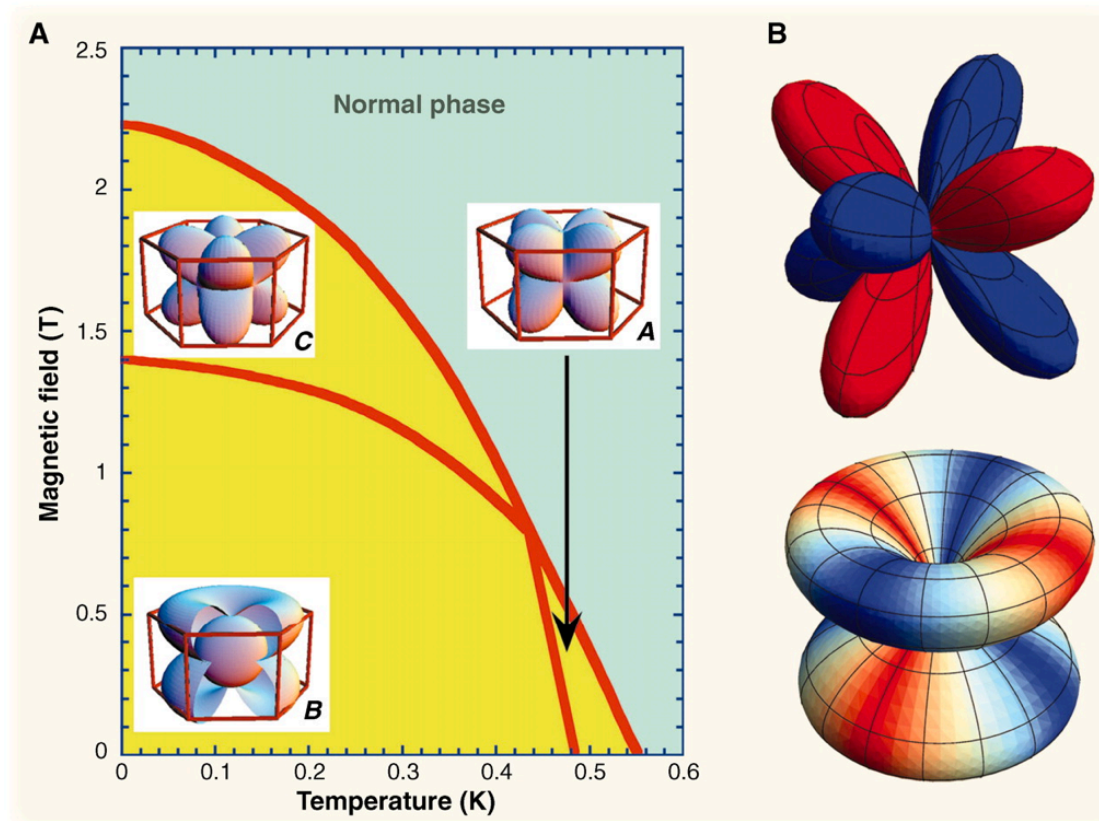
$$\begin{aligned} \Psi_k^{\alpha\beta} = \langle c_{k\alpha} c_{-k\beta} \rangle &= \Psi_s(k) \epsilon_{\alpha\beta} && \text{singlet} \\ &= \Psi_t(k) \cdot (\sigma^y \boldsymbol{\sigma})_{\alpha\beta} && \text{triplet} \end{aligned}$$

- Many varieties of "orbital" state  $\Psi(k)$

s,p,d, p+ip, etc.

# UPt<sub>3</sub>

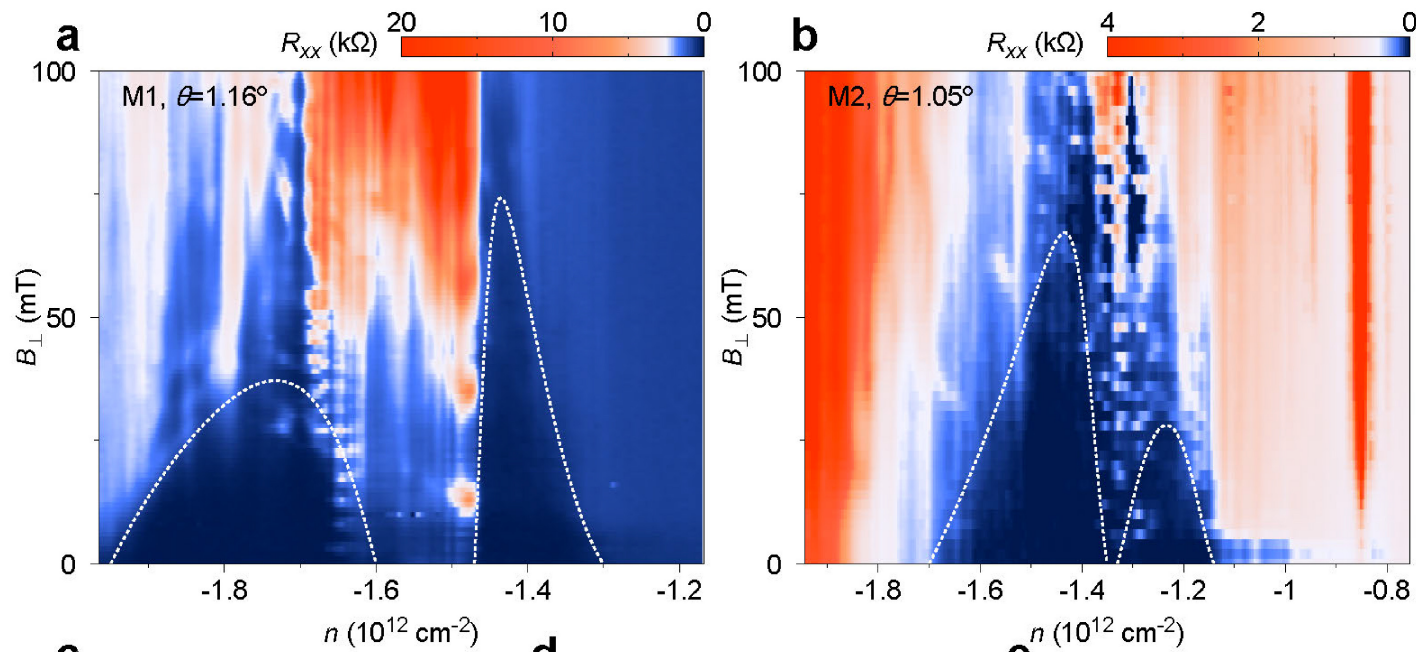
- Several distinct superconducting phases



M. Norman, 2011

# Twisted bilayer graphene

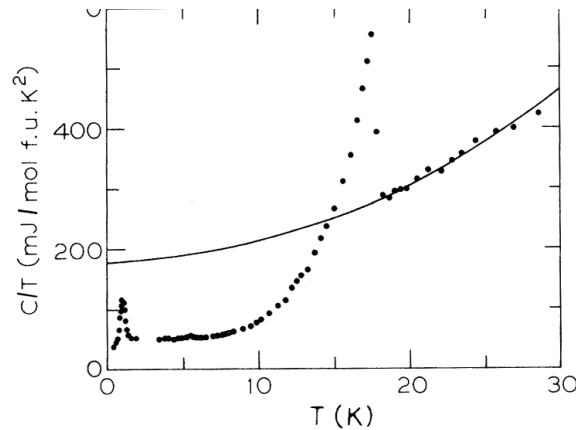
- Recent results (Y. Cao *et al*, 2018)



could be unconventional?

# More subtle order

- Classic example: hidden order



URu<sub>2</sub>Si<sub>2</sub>, Palstra  
*et al*, 1985

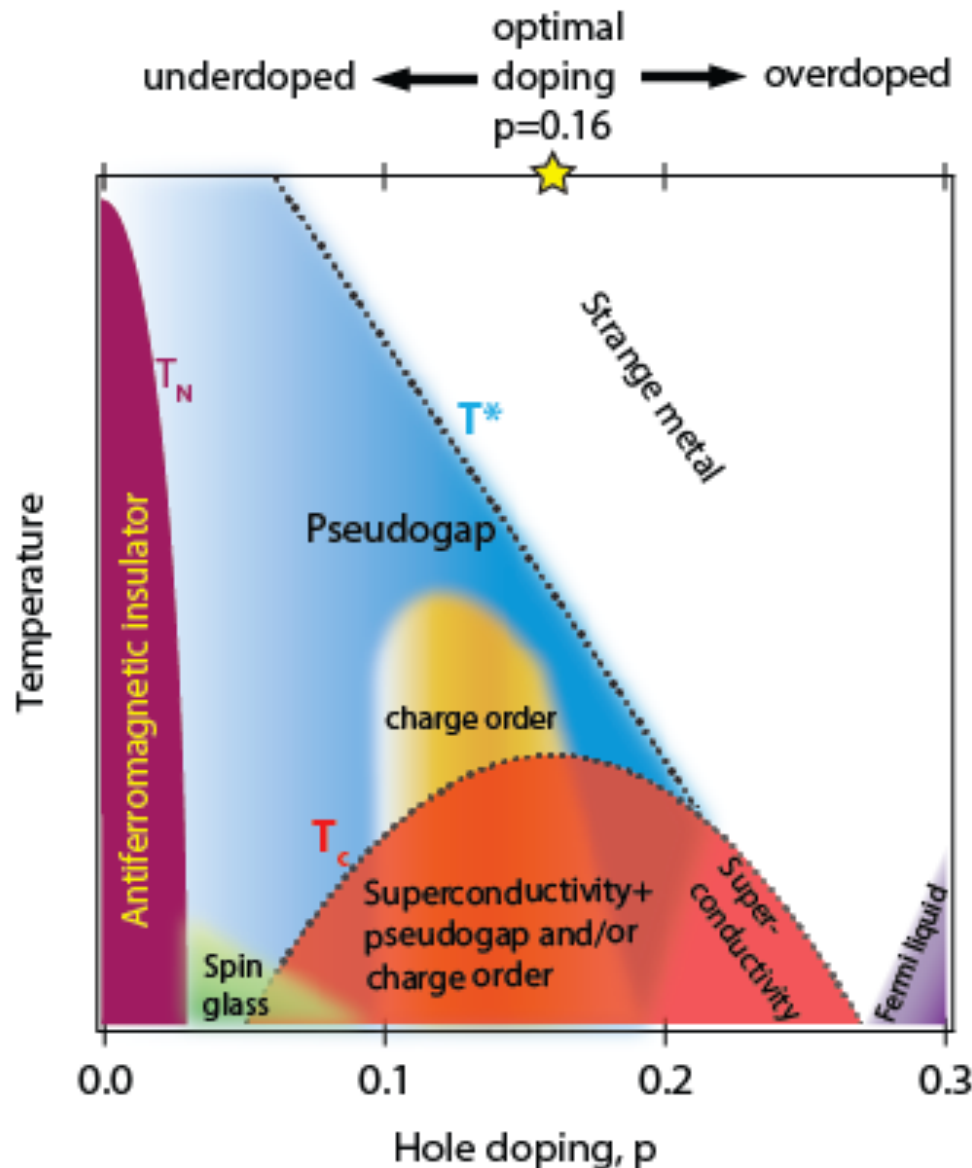
Barzykin and Gorkov (1995)	three-spin correlations [45]
Kasuya (1997)	uranium dimerisation [46]
Ikeda and Ohashi (1998)	<i>d</i> -spin density wave [47]
Okuno and Miyake (1998)	CEF and quantum fluctuations [48]
Chandra et al. (2002)	orbital currents [49]
Viroszek et al. (2002)	unconv. spin density wave [50]
Mineev and Zhitomirsky (2005)	staggered spin density wave [51]
Varma and Zhu (2006)	helicity (Pomeranchuk) order [52]
Elgazzar et al. (2009)	dynamical symmetry breaking [53]
Kotetes et al. (2010)	chiral <i>d</i> -density wave [54]
Dubi and Balatsky (2011)	hybridization wave [55]
Pepin et al. (2011)	modulated spin liquid [56]
Fujimoto (2011)	spin nematic order [57]
Riseborough et al. (2012)	unconv. spin-orbital density wave [58]
Das (2012)	spin-orbital density wave [59]
Chandra et al. (2013)	hastatic order [60]
Hsu and Chakravarty (2013)	singlet-triplet <i>d</i> -density wave [61]

Table 3. Summary of proposals for a specific multipolar magnetic ordering on the uranium ion to explain the HO, with an emphasis on the recent contributions. Note that different symmetries are possible for high-rank multipoles, therefore some kind of multipoles appear more than once.

Nieuwenhuys (1987)	dipole ( $2^1$ ) order [62]
Santini and Amoretti (1994)	quadrupolar ( $2^2$ ) order [63]
Kiss and Fazekas (2005)	octupolar ( $2^3$ ) order [64]
Hanzawa and Watanabe (2005)	octupolar order [65]
Hanzawa (2007)	incommensurate octupole [66]
Haule and Kotliar (2009)	hexadecapolar ( $2^4$ ) order [67]
Cricchio et al. (2009)	dotriacontapolar ( $2^5$ ) order [68]
Harima et al. (2010)	antiferro quadrupolar order [69]
Thalmeier and Takimoto (2011)	<i>E</i> (1, 1)-type quadrupole [70]
Kusunose and Harima (2011)	antiferro hexadecapole [71]
Ikeda et al. (2012)	<i>E</i> <sup>-</sup> -type dotriacontapole [72]
Rau and Kee (2012)	<i>E</i> -type dotriacontapole [73]
Ressouche et al. (2012)	dotriacontapolar order [16]

from Mydosh + Openeer, 2014

# Pseudo-gap



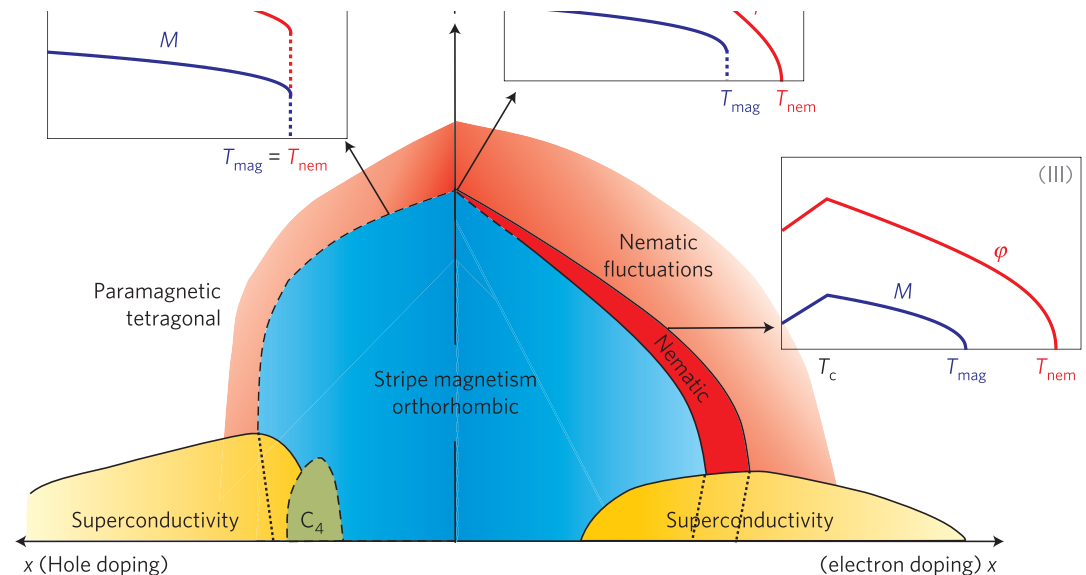
charge order  
loop currents  
T-reversal breaking  
pair density wave  
stripes  
nematic  
RVB

...

competing/intertwined/  
hidden/fluctuating orders??

# Electron nematic

*symmetry* is just tetragonal to orthorhombic, but seems to be that electrons, not lattice, are driving the order



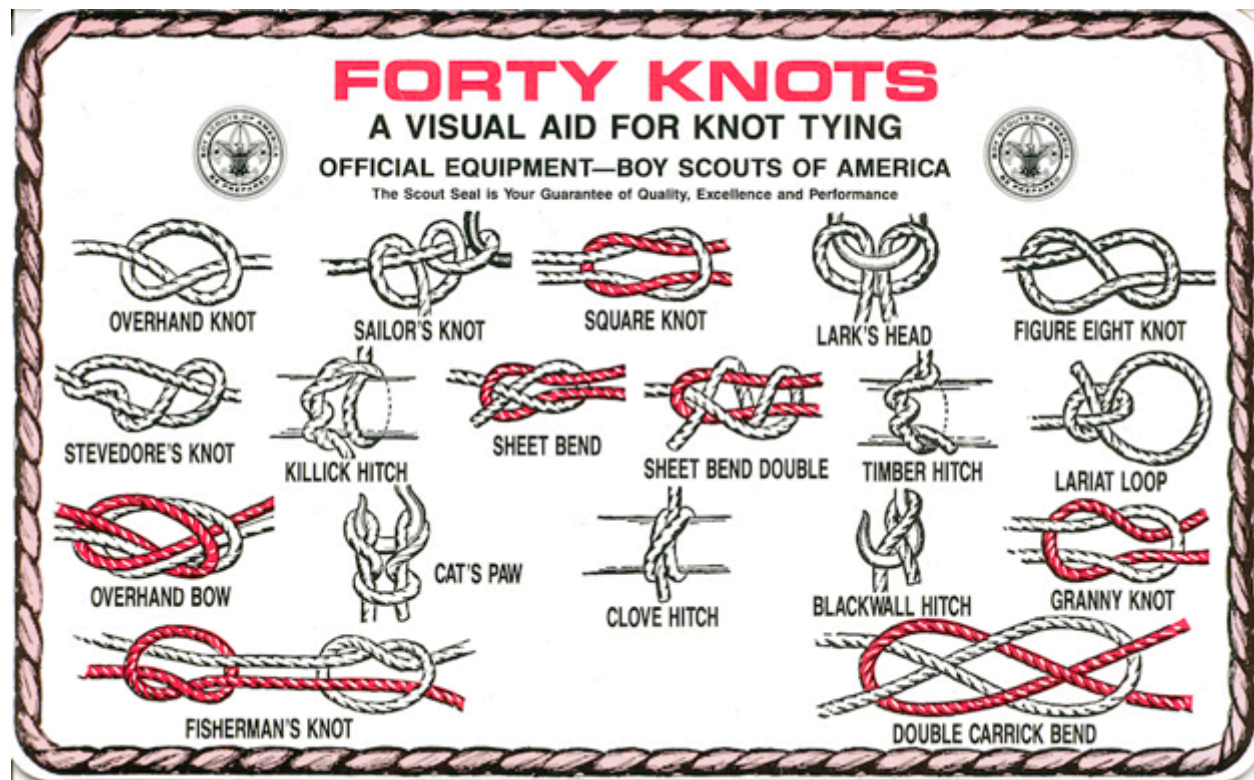
schematic of Fe122 materials from  
Fernandez et al, 2014

# Order questions

- What is the hidden order in  $\text{URu}_2\text{Si}_2$ ?
- Is the pseudo-gap region of the cuprates a consequence of some hidden order?
- Are there mechanisms for unifying different orders in QMs - i.e. can you give a sharp meaning to intertwinement etc.?

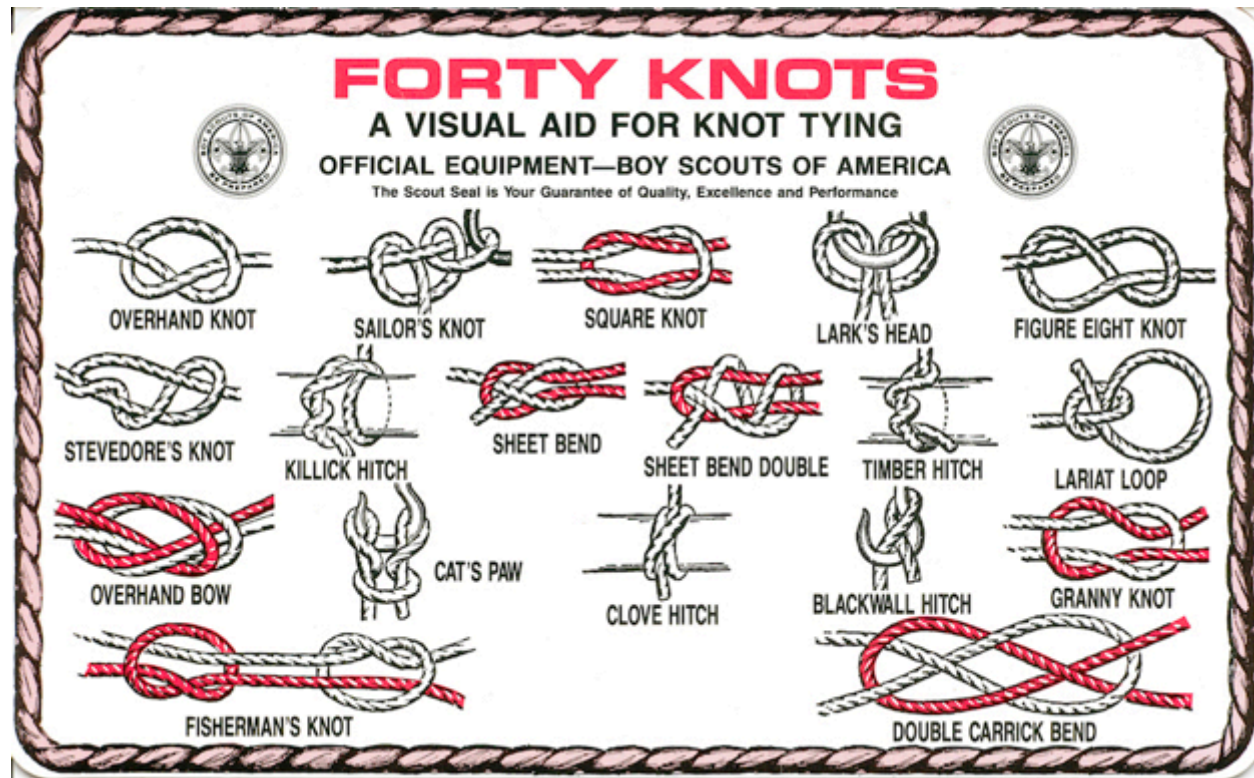
# Topology

Mathematical study of objects that can and cannot be smoothly transformed into one another

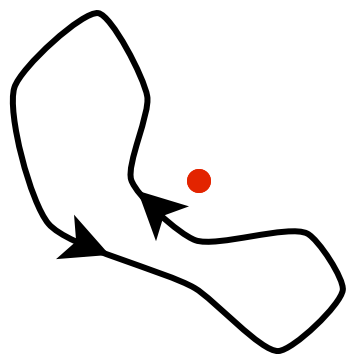


# Topology

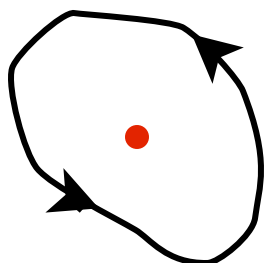
In physics, “knotted” states can become new phases of matter with robust properties



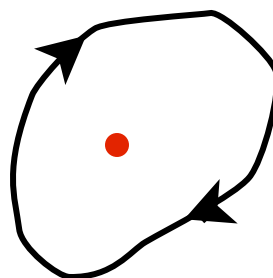
# Topological invariant



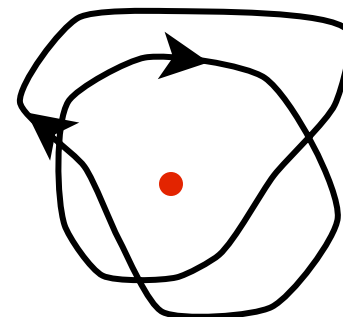
$$W=0$$



$$W=1$$



$$W=-1$$



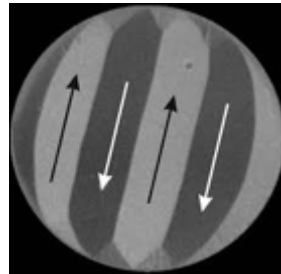
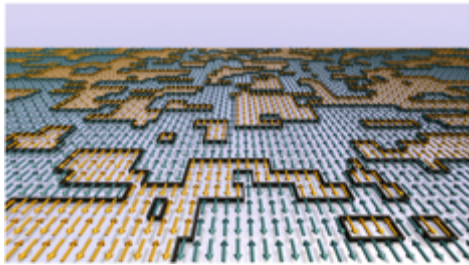
$$W=-2$$

Winding number

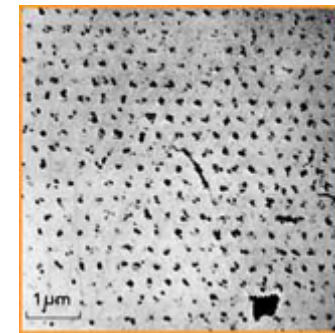
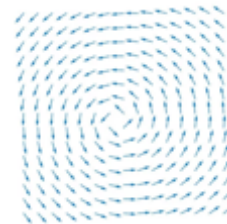
$$\begin{aligned} W[\mathcal{C}] &= \frac{1}{2\pi i} \oint_{\mathcal{C}} \frac{dz}{z} \in \mathbb{Z} \\ &= \frac{1}{2\pi i} \int ds \frac{z'(s)}{z(s)} \end{aligned}$$

# Topology of order

- Spontaneously broken symmetry leads to topological defects: configurations of order parameter that cannot be smoothly “unwound”



discrete order: domain walls



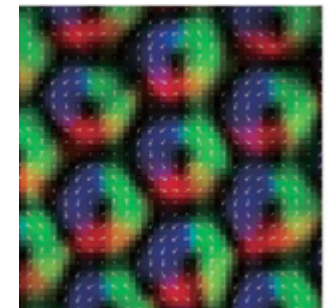
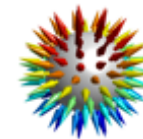
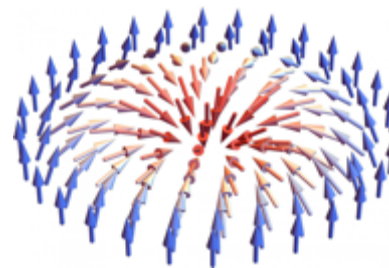
XY/U(1) order: vortices

## The topological theory of defects in ordered media<sup>\*†</sup>

N. D. Mermin

*Laboratory of Atomic and Solid State Physics Cornell University, Ithaca, New York 14853*

Aspects of the theory of homotopy groups are described in a mathematical style closer to that of condensed matter physics than that of topology. The aim is to make more readily accessible to physicists the recent applications of homotopy theory to the study of defects in ordered media. Although many physical examples are woven into the development of the subject, the focus is on mathematical pedagogy rather than on a systematic review of applications.



vector order: skyrmions/hedgehogs



# Topological band theory

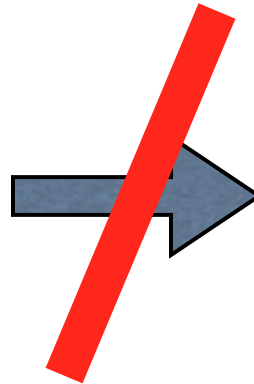
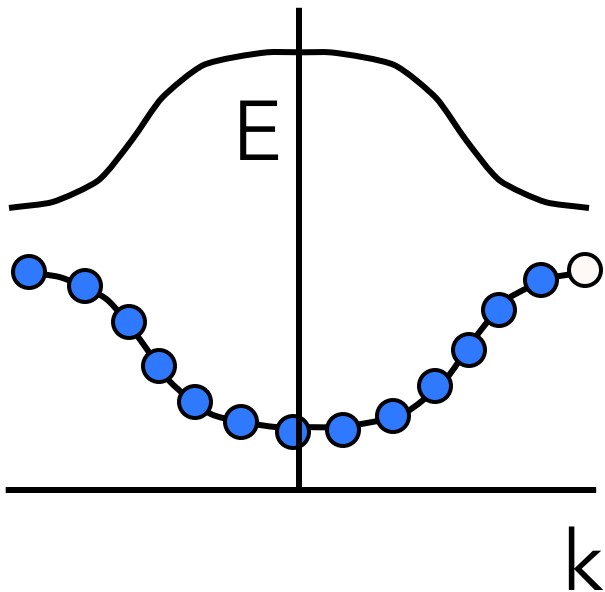


## Über die Quantenmechanik der Elektronen in Kristallgittern.

Von **Felix Bloch** in Leipzig.

Mit 2 Abbildungen. (Eingegangen am 10. August 1928.)

Die Bewegung eines Elektrons im Gitter wird untersucht, indem wir uns dieses durch ein zunächst streng dreifach periodisches Kraftfeld schematisieren. Unter Hinzunahme der Fermischen Statistik auf die Elektronen gestattet unser Modell Aussagen über den von ihnen herrührenden Anteil der spezifischen Wärme des Kristalls. Ferner wird gezeigt, daß die Berücksichtigung der thermischen Gitterschwingungen Größenordnung und Temperaturabhängigkeit der elektrischen Leitfähigkeit von Metallen in qualitativer Übereinstimmung mit der Erfahrung ergibt.



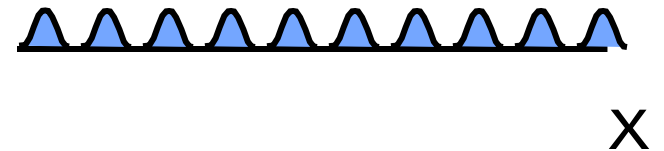
Thouless, 1984

## The Structure of Electronic Excitation Levels in Insulating Crystals

GREGORY H. WANNIER  
*Princeton University, Princeton, New Jersey\**

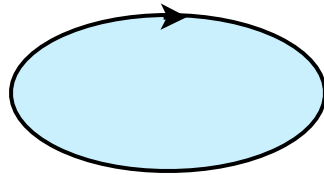
(Received May 13, 1937)

In this article, a method is devised to study the energy spectrum for an excited electron configuration in an ideal crystal. The configuration studied consists of a single excited electron taken out of a full band of  $N$  electrons. The multiplicity of the state is  $N^2$ . It is shown that because of the Coulomb attraction between the electron and its hole  $N^{3/2}$  states are split off from the bottom of the excited Bloch band; for these states the electron cannot escape its hole completely. The analogy of these levels to the spectrum of an atom or molecule is worked out quantitatively. The bottom of the Bloch band appears as "ionization potential" and the Bloch band itself as the continuum above this threshold energy.



# Chern number

- Bloch states  $\psi_n(r) = e^{ik \cdot r} u_{n,k}(r)$
- Berry gauge field  $\vec{\mathcal{A}}_n = i \langle u_n | \vec{\nabla}_k | u_n \rangle$

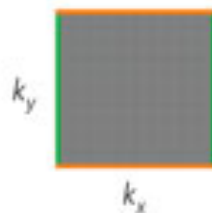


$$\vec{\mathcal{B}}_n = \vec{\nabla}_k \times \vec{\mathcal{A}}_n$$

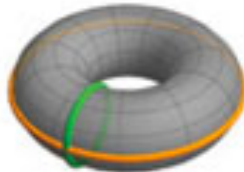
- Net Berry flux gives Chern number

$$q_n = \frac{1}{2\pi} \int d^2k \mathcal{B}_n^z \in \mathbb{Z}$$

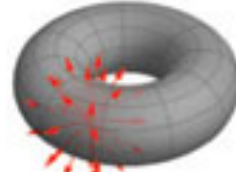
2D Brillouin zone



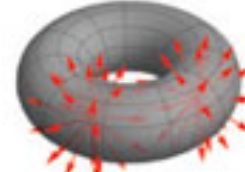
$C = 0$



$C = 1$

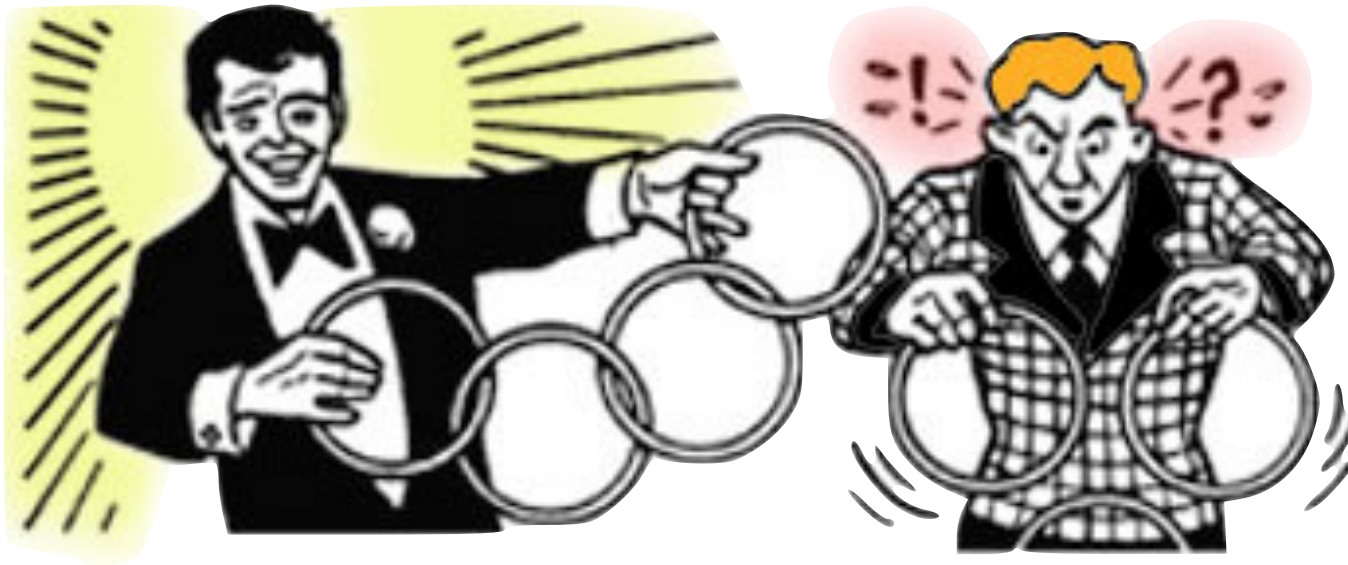


$C = 2$



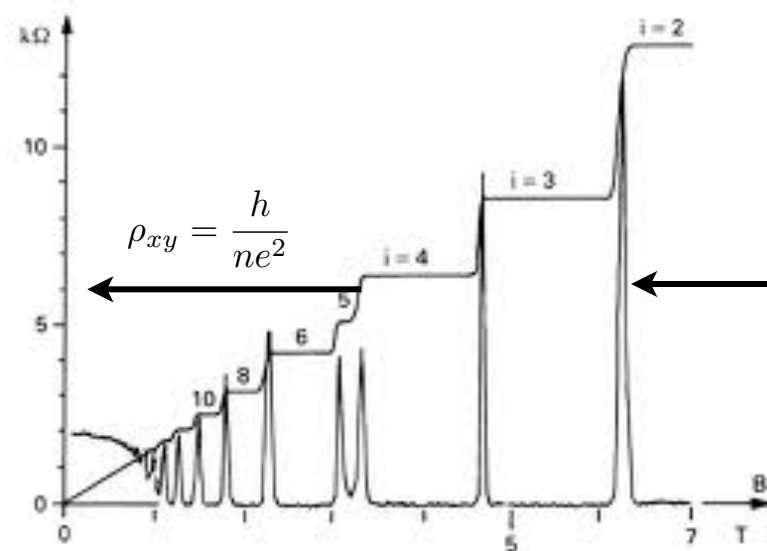
# Phase transitions

- There is no *smooth* way to go from one topological sector to another

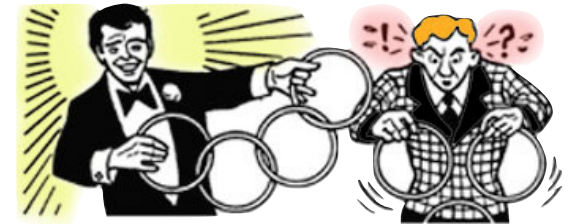


# IQHE

- Hall conductance measures Chern number



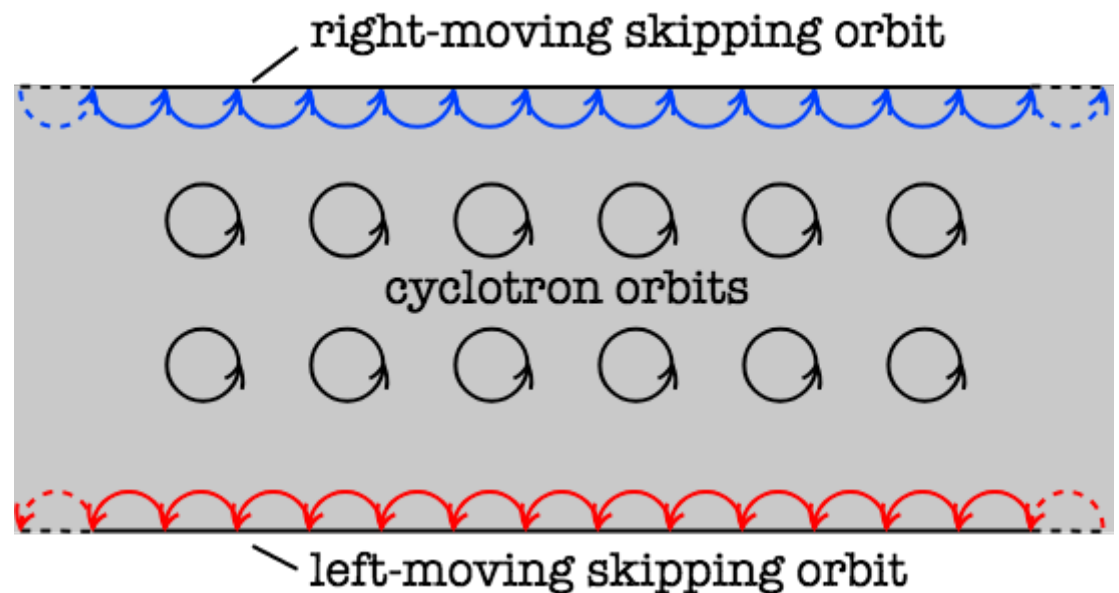
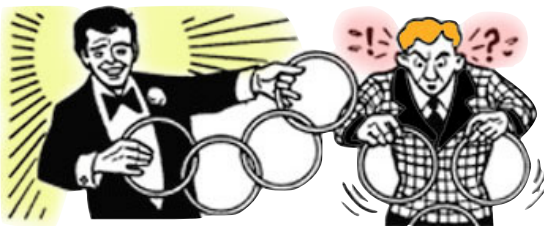
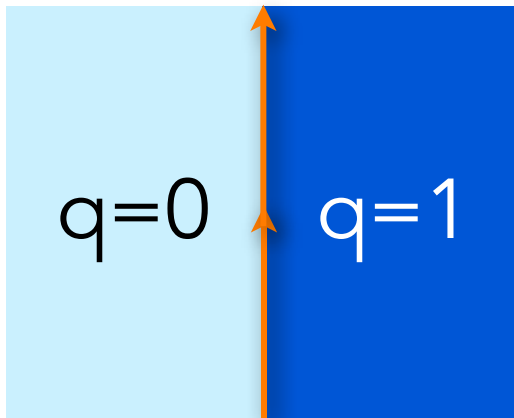
$\rho_{xx}$  peak signifies gapless quantum critical state



- Passage from one Chern number to another is a quantum phase transition  
(with no order parameter!)

# IQHE

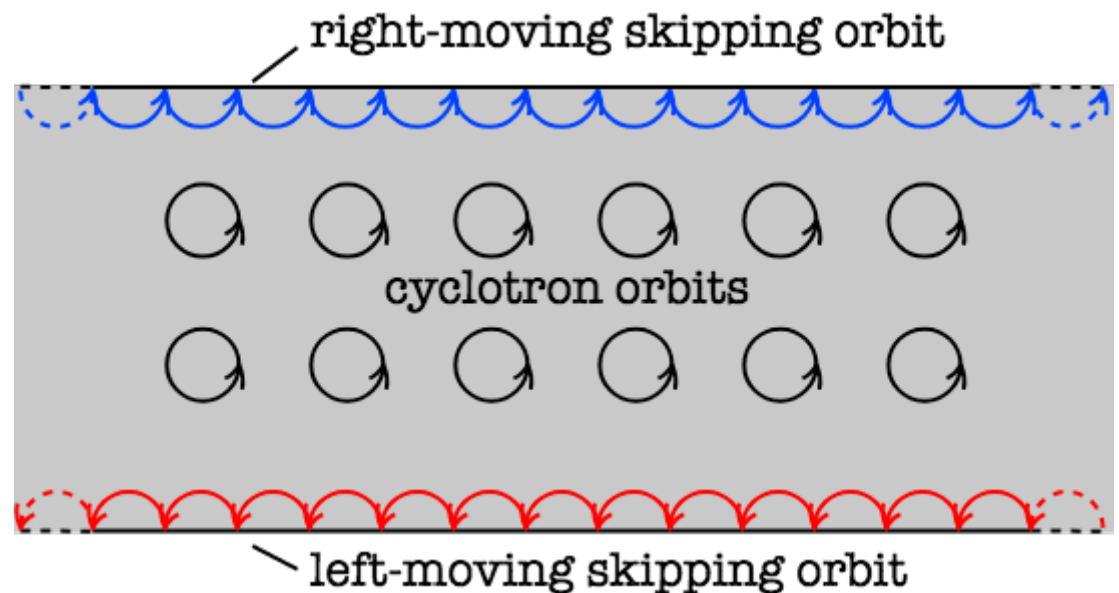
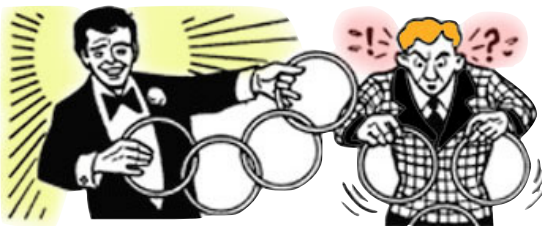
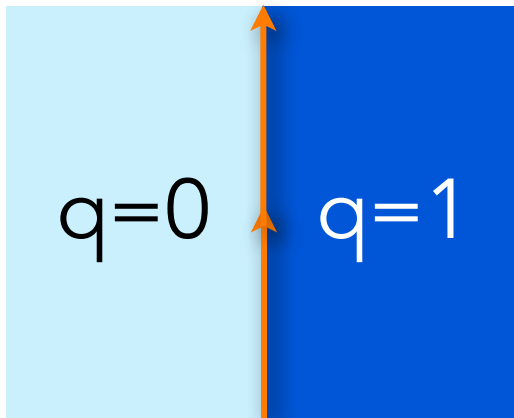
- The boundary between states with different Chern number is gapless



*chiral* edge states cannot backscatter

# IQHE

- The boundary between states with different Chern number is gapless

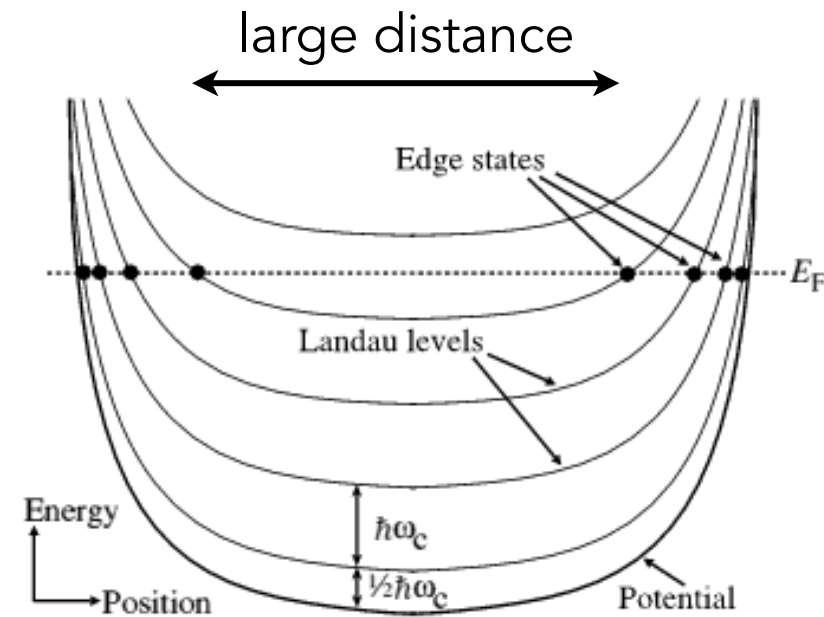
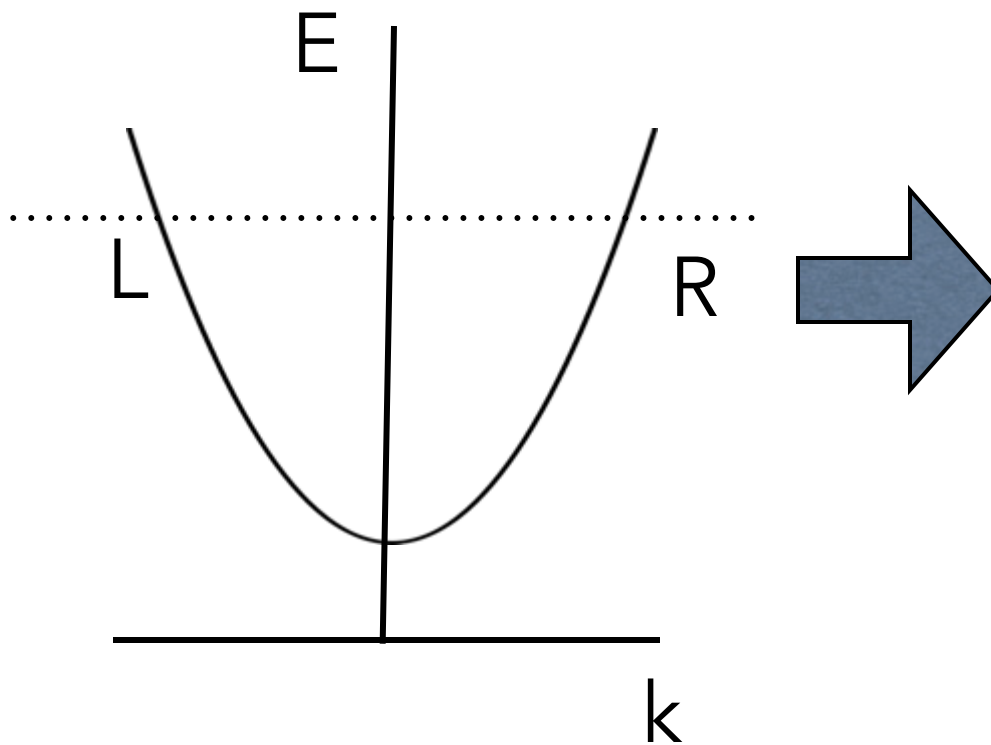


Bulk-boundary correspondence:

$$\text{Chern Number} = N_R - N_L$$

# IQHE

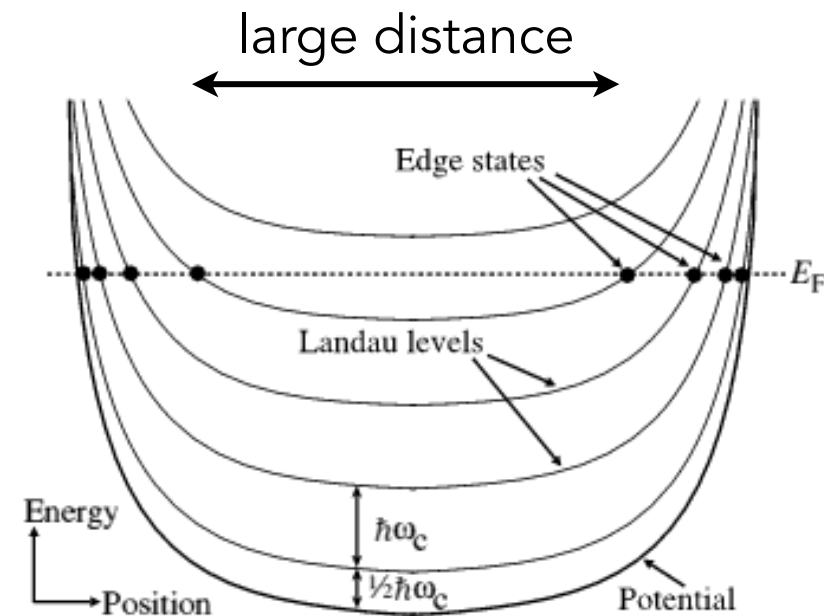
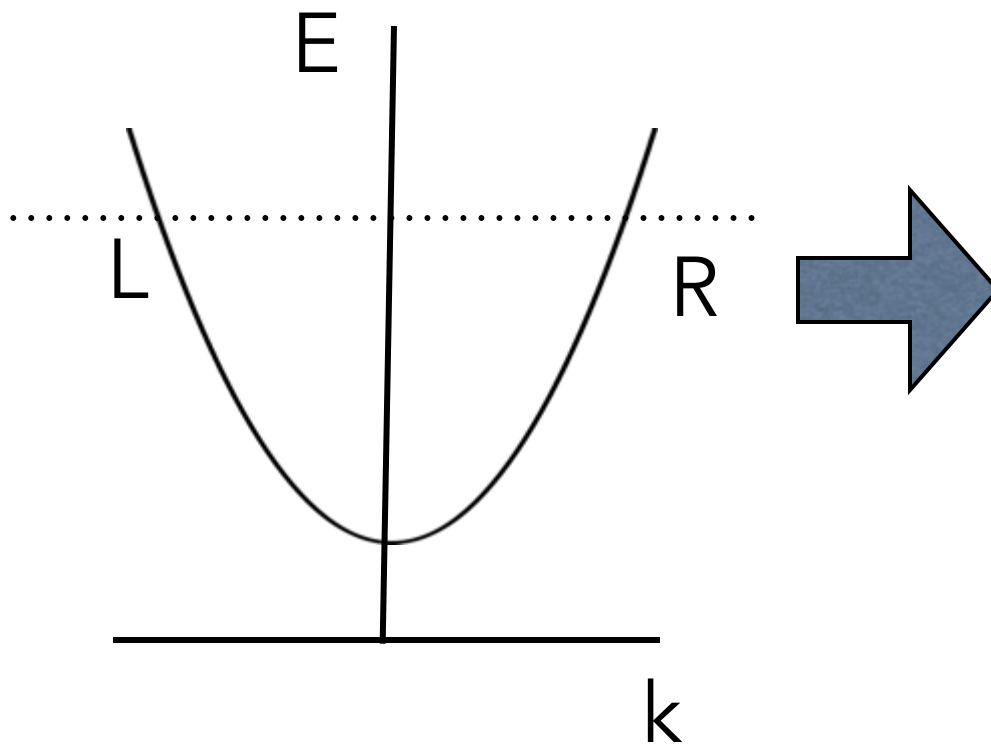
- Edge states are “half” of the low energy excitations of a 1DEG



Halperin, 1982

# IQHE

- Edge states are “half” of the low energy excitations of a 1DEG



Halperin, 1982

General rule: surface state of a  $d$ -dimensional TI cannot be realized in a  $d-1$  dimensional system

# Summary - Chern Insulators

- Winding of the one-electron wavefunction over the Brillouin zone
- Quantified by a topological invariant: the Chern number, which coincides with Hall conductance
- States with different values of the topological invariant are different phases
- Tuning from one to another requires a quantum phase transition
- The interface between two different values has gapless “protected” edge states, which are “anomalous”: they could not exist in an isolated 1-dimensional system

# Time-reversal symmetry

- Would be nice to have topological quantization in materials in “natural conditions”
- The Berry curvature is odd under time-reversal  $\mathcal{B}(-k) = -\mathcal{B}(k)$
- This implies the Chern number vanishes
- For decades it was believed this meant electronic states are topologically trivial with TRS

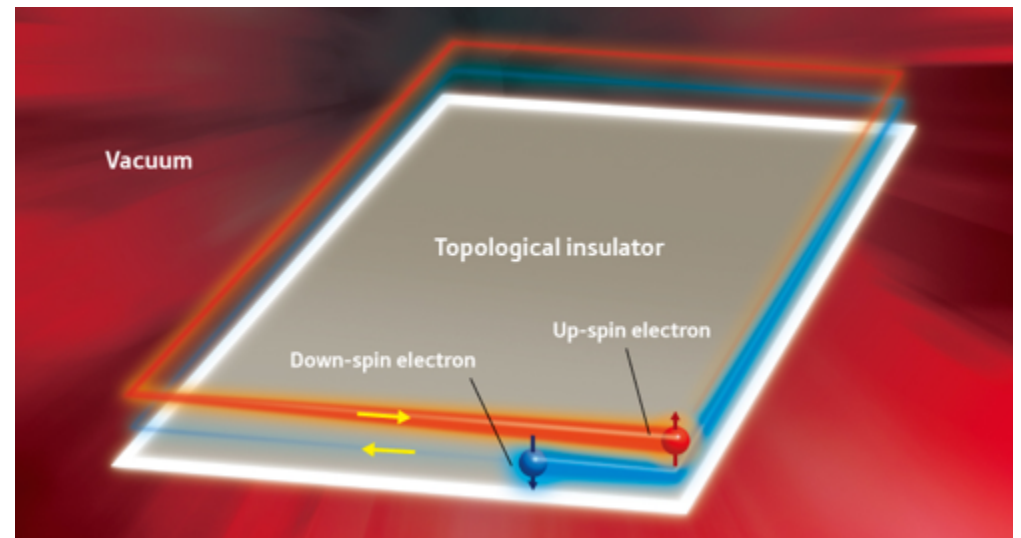


# $Z_2$ TIs

2d: Kane, Mele (2005); Bernevig, Hughes, Zhang (2006)

3d: L. Fu, C. Kane, E. Mele (2007); J. Moore, LB (2007)

- Even with TR symmetry, a different type of TI is possible (*with spin-orbit coupling*)
- 2d: “QSHE”
- Roughly understood as opposite IQHE’s for up and down electrons

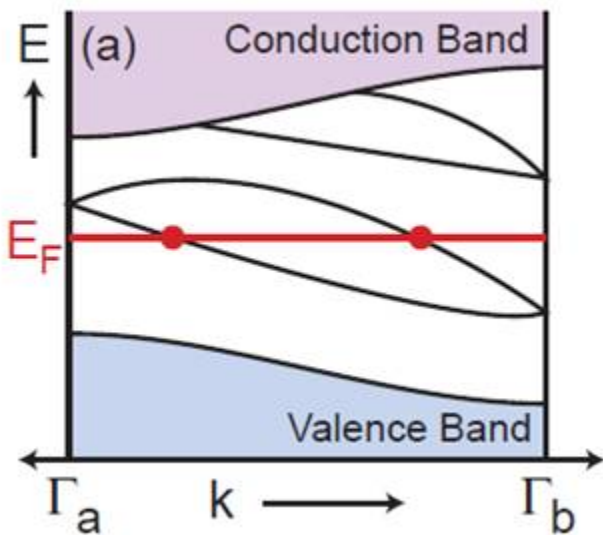


# $Z_2$ TIs

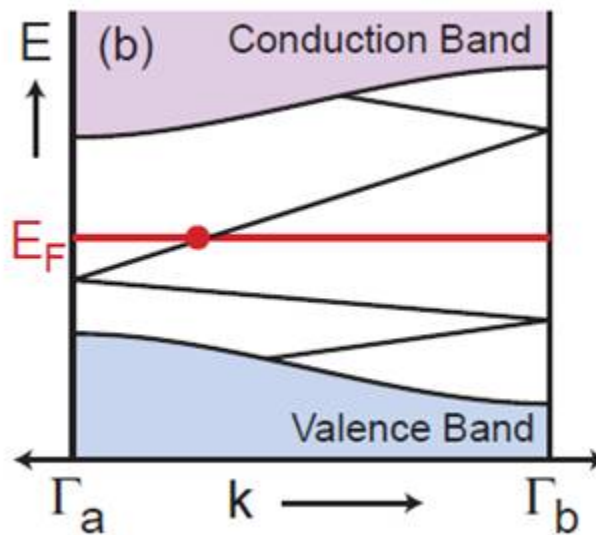
2d: Kane, Mele (2005); Bernevig, Hughes, Zhang (2006)

3d: J. Moore, LB (2007); L. Fu, C. Kane, E. Mele (2007)

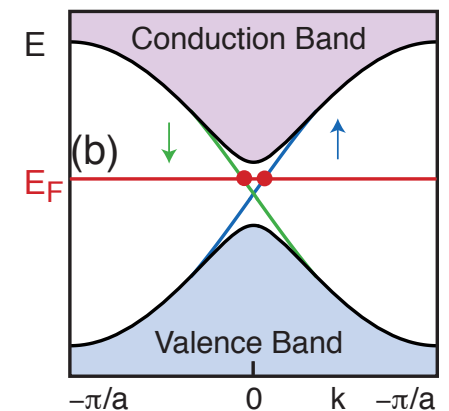
- $Q$  = parity of band crossings between TRI momenta at the surface



"trivial"



"topological"



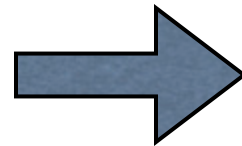
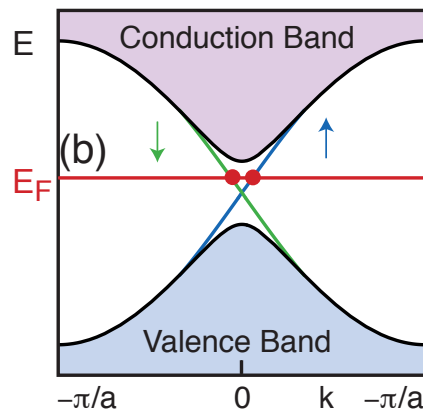
Full zone for TI

# $Z_2$ TIs

J. Moore, LB (2007)

L. Fu, C. Kane, E. Mele (2007)

- In 3d, there are *four*  $Z_2$  parities
  - 3 “weak” parities - describe layered 2d TIs
  - 1 “strong” parity - describes intrinsically 3d physics

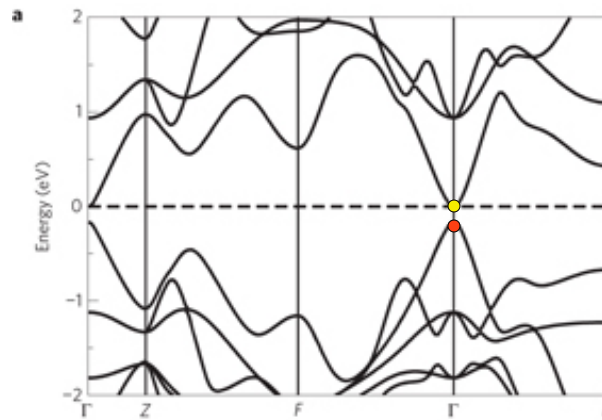


odd number  
of Dirac  
cones at any  
surface

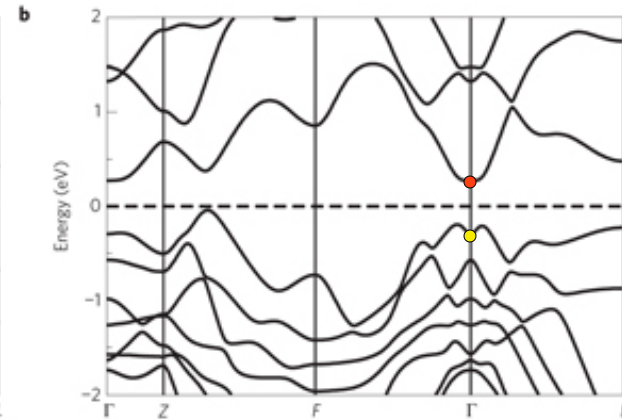
# Band inversion

- Different classes of bands cannot be smoothly deformed into one another

$\text{Bi}_2\text{Se}_3$



no SOC

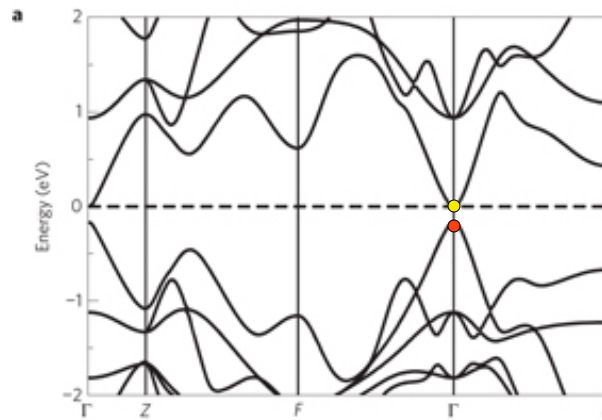


w/ SOC

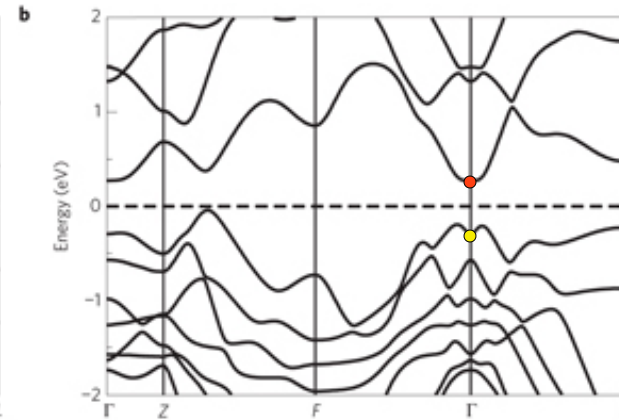
# Band inversion

- Different classes of bands cannot be smoothly deformed into one another

$\text{Bi}_2\text{Se}_3$



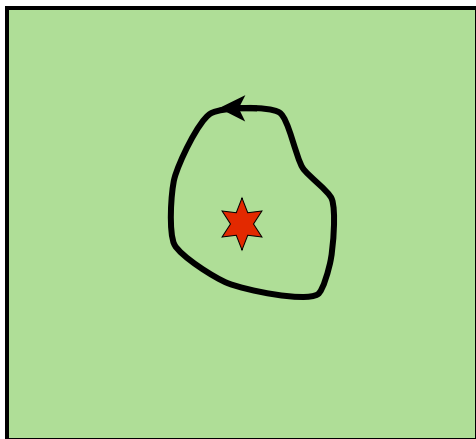
normal insulator



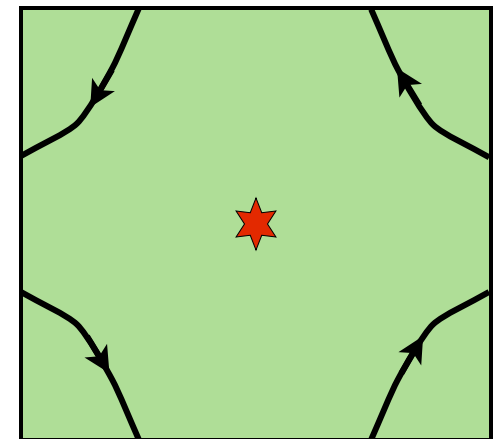
topological insulator

# Single Dirac cone

- 1/4 Graphene
- 2d Dirac mass breaks TR
- Cannot be found in any 2d model



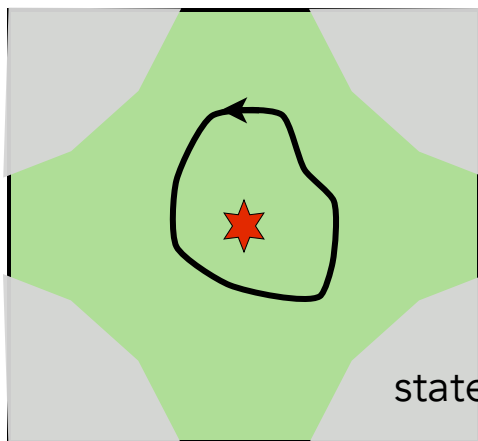
$$\Phi = \pi$$



$$\Phi = 0$$

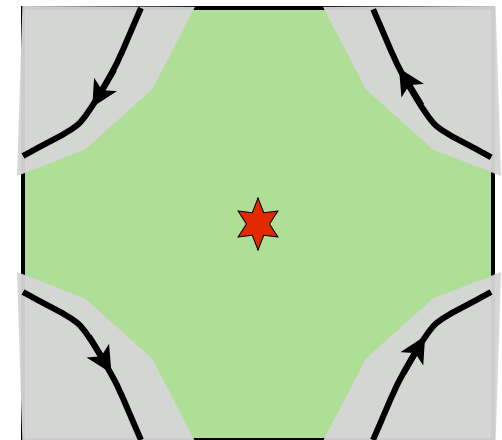
# Single Dirac cone

- 1/4 Graphene
- 2d Dirac mass breaks TR
- Cannot be found in any 2d model



states decay into bulk

$$\Phi = \pi$$



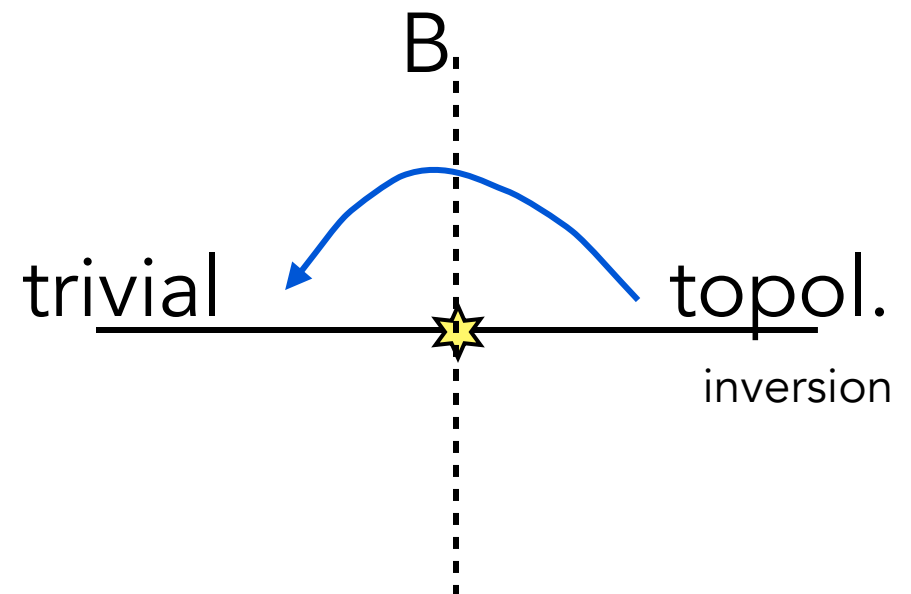
not allowed

# Single Dirac cone

- 1/4 Graphene
- 2d Dirac mass breaks TR
- Cannot be found in any 2d model



If time reversal is broken, the Dirac cone is gapped and TI becomes equivalent to trivial insulator



# TIs

Broken  
TRS

integer Chern  
number

= Hall  
conductance

chiral edge  
states

with  
TRS

$Z_2$  invariant

= magneto-  
electric  
polarizability

helical edge/  
surface states

# Topological superconductors

“periodic table” of TIs+TSCs

Chern insulator	$Z_2$ TI
chiral topo. SC	$Z_2$ Topo. SC

He3B

free fermion

TIs

Name	T	C	S=CT	d=1	d=2	d=3	
A	0	0	0	-	$\mathbb{Z}$	-	IQHE
AIII	0	0	1	$\mathbb{Z}$	-	$\mathbb{Z}$	
AI	+1	0	0	-	-	-	polyacetylen
BDI	+1	+1	1	$\mathbb{Z}$	-	-	chiral p-wave
D	0	+1	0	$\mathbb{Z}_2$	$\mathbb{Z}$	-	TRI top. triplet SC (He3 B)
DIII	-1	+1	1	$\mathbb{Z}_2$	$\mathbb{Z}_2$	$\mathbb{Z}$	QSHE
AII	-1	0	0	-	$\mathbb{Z}_2$	$\mathbb{Z}_2$	3D $Z_2$ top. insulator
CII	-1	-1	1	$\mathbb{Z}$	-	$\mathbb{Z}_2$	chiral d-wave
C	0	-1	0	-	$\mathbb{Z}$	-	
CI	+1	-1	1	-	-	$\mathbb{Z}$	TRI top. singlet SC

A. Schnyder  
*et al*, 2008

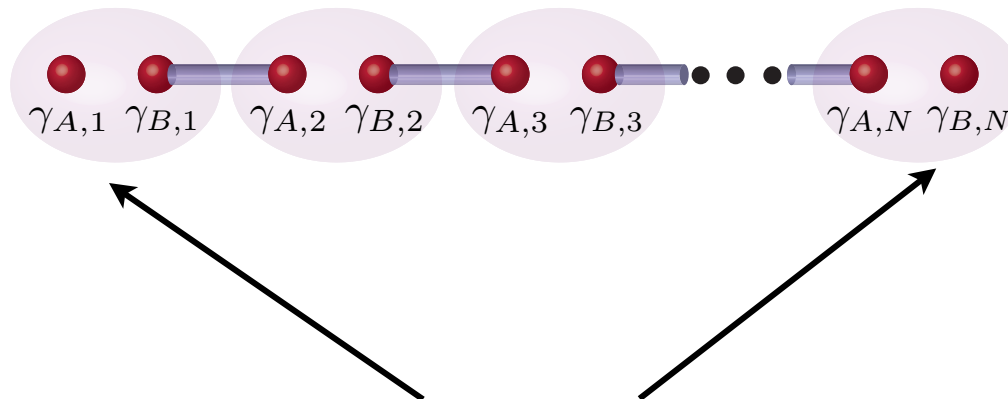
Kitaev, 2009

# Kitaev Chain

- 1d spinless “p-wave” superconductor

$$H = -\mu \sum_{x=1}^N c_x^\dagger c_x - \sum_{x=1}^{N-1} (t c_x^\dagger c_{x+1} + |\Delta| e^{i\phi} c_x c_{x+1} + h.c.)$$

- Topological for  $|\mu| < 2t$



together these two “Majorana”  
fermions form *one* two-level system

# Topological semi-metals

- Another class of topological states are *semi-metals*, in which band touching is protected by topology
  - Graphene: 2d Dirac fermion
  - TaAs: 3d Weyl fermion
  - $\text{Cd}_3\text{Se}_2$ : 3d Dirac fermion
  - $\text{Mn}_3\text{Sn}$ : 3d magnetic Weyl fermions
  - many more!
- In these systems the touching points are like “topological defects”: singularities of Berry curvature
- Near the touching points, electrons have unusual dynamics: lots of potential for interesting physics!

# Weyl semimetal

For a crystal without an inversion center, the energy separation  $\delta E(\mathbf{k}+\boldsymbol{\kappa})$  in the neighborhood of a point  $\mathbf{k}$  where contact of equivalent manifolds occurs may be expected to be of the order of  $\kappa$  as  $\kappa \rightarrow 0$ , for all directions of  $\boldsymbol{\kappa}$ .

937

PHYSICAL REVIEW

Accidental Degeneracy in the Energy Bands of Crystals

CONYERS HERRING

*Princeton University, Princeton, New Jersey*

(Received June 16, 1937)

# Weyl semimetal

937

PHYSICAL REVIEW

## Accidental Degeneracy in the Energy Bands of Crystals

CONYERS HERRING

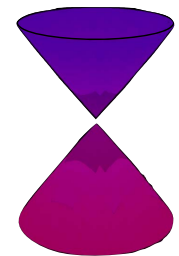
Princeton University, Princeton, New Jersey

(Received June 16, 1937)



For a crystal without an inversion center, the energy separation  $\delta E(\mathbf{k}+\boldsymbol{\kappa})$  in the neighborhood of a point  $\mathbf{k}$  where contact of equivalent manifolds occurs may be expected to be of the order of  $\kappa$  as  $\kappa \rightarrow 0$ , for all directions of  $\boldsymbol{\kappa}$ .

$$H = v \vec{\sigma} \cdot \vec{k}$$



A two-component spinor in three dimensions: "half" of a Dirac fermion. Weyl fermions have a chirality and *must* be massless

(Dirac semimetals also exist)

# Weyl semimetal

For a crystal without an inversion center, the energy separation  $\delta E(\mathbf{k}+\boldsymbol{\kappa})$  in the neighborhood of a point  $\mathbf{k}$  where contact of equivalent manifolds occurs may be expected to be of the order of  $\kappa$  as  $\kappa \rightarrow 0$ , for all directions of  $\boldsymbol{\kappa}$ .

937

PHYSICAL REVIEW

Accidental Degeneracy in the Energy Bands of Crystals

CONYERS HERRING

*Princeton University, Princeton, New Jersey*

(Received June 16, 1937)

Either inversion or time-reversal  
(or both) must be broken

# Weyl semimetal

For a crystal without an inversion center, the energy separation  $\delta E(\mathbf{k}+\boldsymbol{\kappa})$  in the neighborhood of a point  $\mathbf{k}$  where contact of equivalent manifolds occurs may be expected to be of the order of  $\kappa$  as  $\kappa \rightarrow 0$ , for all directions of  $\boldsymbol{\kappa}$ .

937

PHYSICAL REVIEW

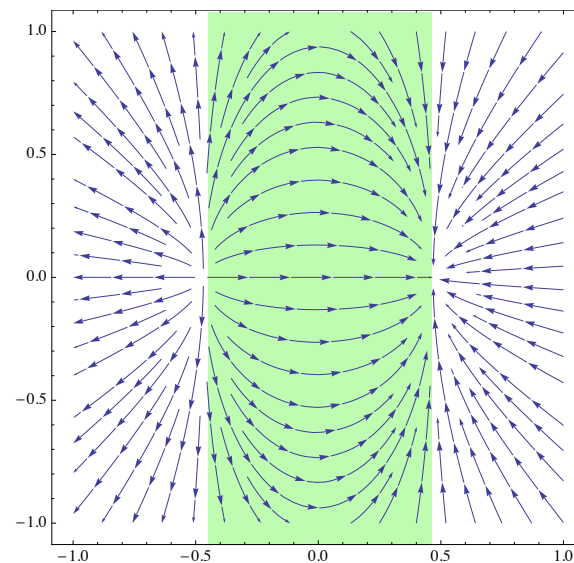
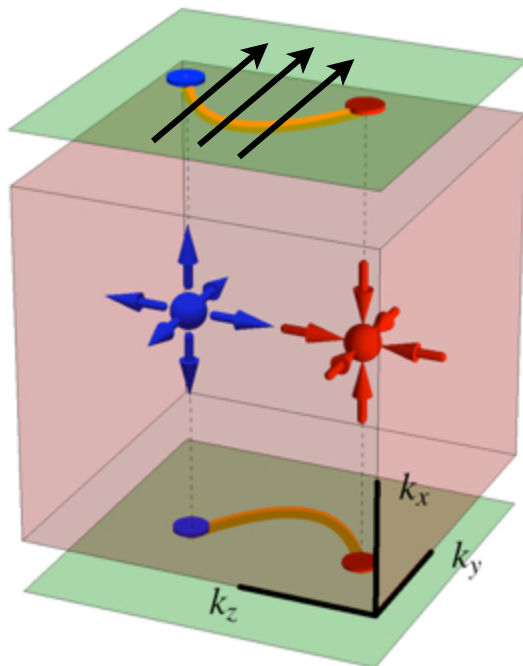
## Accidental Degeneracy in the Energy Bands of Crystals

CONYERS HERRING  
Princeton University, Princeton, New Jersey  
(Received June 16, 1937)

S. Murakami, 2007

X. Wan et al, 2011

A. Burkov+LB, 2011



Weyl points are  
“monopoles” of  
Berry curvature:  
topology in  $k$ -  
space!

Review: A.M. Turner, A. Vishwanath, arXiv:1301.0330

# Weyl semimetal

For a crystal without an inversion center, the energy separation  $\delta E(\mathbf{k}+\boldsymbol{\kappa})$  in the neighborhood of a point  $\mathbf{k}$  where contact of equivalent manifolds occurs may be expected to be of the order of  $\kappa$  as  $\kappa \rightarrow 0$ , for all directions of  $\boldsymbol{\kappa}$ .

937

PHYSICAL REVIEW

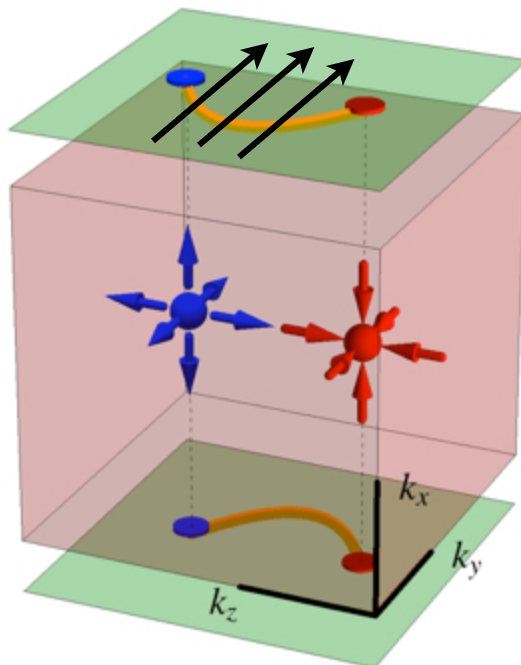
## Accidental Degeneracy in the Energy Bands of Crystals

CONYERS HERRING  
Princeton University, Princeton, New Jersey  
(Received June 16, 1937)

S. Murakami, 2007

X. Wan et al, 2011

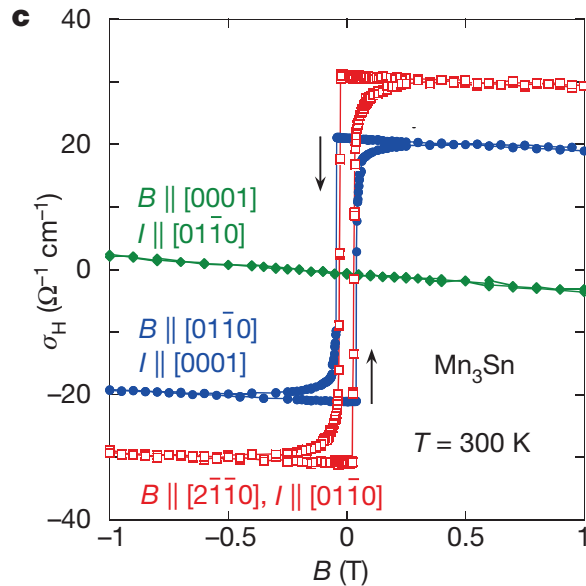
A. Burkov+LB, 2011



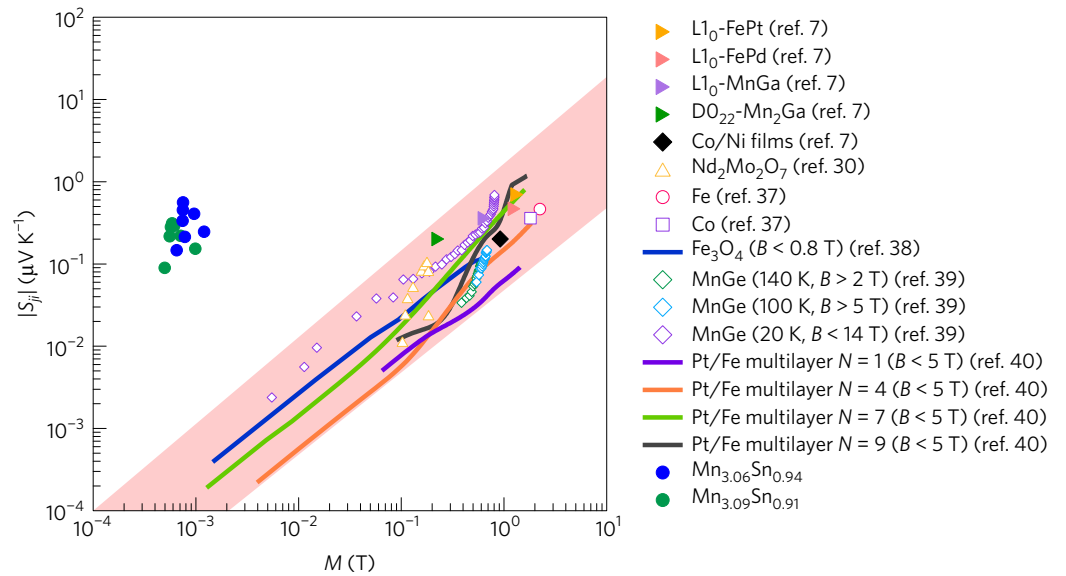
- Striking properties:
  - Anomalous Hall effect
  - ABJ "anomaly": strong negative MR for  $I \parallel B$
  - Surface Fermi arcs

Review: A.M. Turner, A. Vishwanath, arXiv:1301.0330

# Mn<sub>3</sub>Sn magnetic Weyl



anomalous Hall effect,  
Nakatsuji et al, 2015



anomalous Nernst effect,  
Ikhlas et al, 2017

Effects comparable to or exceeding  
ferromagnetic metals at room temperature

# Topological band theory

- Lots more topology if you include all symmetries and atomic structure: “TCIs”

ARTICLE

DOI: 10.1038/s41467-017-00133-2

OPEN

## Symmetry-based indicators of band topology in the 230 space groups

Hoi Chun Po<sup>1,2</sup>, Ashvin Vishwanath<sup>1,2</sup> & Haruki Watanabe<sup>3</sup>

The interplay between symmetry and topology leads to a rich variety of electronic topological phases, protecting states such as the topological insulators and Dirac semimetals. Previous results, like the Fu-Kane parity criterion for inversion-symmetric topological insulators, demonstrate that symmetry labels can sometimes unambiguously indicate underlying band topology. Here we develop a systematic approach to expose all such symmetry-based indicators of band topology in all the 230 space groups. This is achieved by first developing an efficient way to represent band structures in terms of elementary basis states, and then isolating the topological ones by removing the subset of atomic insulators, defined by the existence of localized symmetric Wannier functions. Aside from encompassing all earlier results on such indicators, including in particular the notion of filling-enforced quantum band insulators, our theory identifies symmetry settings with previously hidden forms of band topology, and can be applied to the search for topological materials.

## Topological quantum chemistry

Barry Bradlyn<sup>1\*</sup>, L. Elcoro<sup>2\*</sup>, Jennifer Cano<sup>1\*</sup>, M. G. Vergniory<sup>3,4,5\*</sup>, Zhijun Wang<sup>6\*</sup>, C. Felser<sup>7</sup>, M. I. Aroyo<sup>2</sup> & B. Andrei Bernevig<sup>3,6,8,9</sup>

Since the discovery of topological insulators and semimetals, there has been much research into predicting and experimentally discovering distinct classes of these materials, in which the topology of electronic states leads to robust surface states and electromagnetic responses. This apparent success, however, masks a fundamental shortcoming: topological insulators represent only a few hundred of the 200,000 stoichiometric compounds in material databases. However, it is unclear whether this low number is indicative of the esoteric nature of topological insulators or of a fundamental problem with the current approaches to finding them. Here we propose a complete electronic band theory, which builds on the conventional band theory of electrons, highlighting the link between the topology and local chemical bonding. This theory of topological quantum chemistry provides a description of the universal (across materials), global properties of all possible band structures and (weakly correlated) materials, consisting of a graph-theoretic description of momentum (reciprocal) space and a complementary group-theoretic description in real space. For all 230 crystal symmetry groups, we classify the possible band structures that arise from local atomic orbitals, and show which are topologically non-trivial. Our electronic band theory sheds new light on known topological insulators, and can be used to predict many more.

Day 2

# Topological frontiers

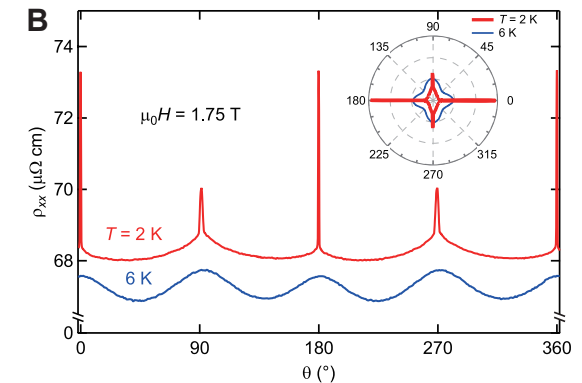
- Can we realize a topological superconductor?
- What is the interplay between topological defects and topological bands?
- What are the robust signatures of topology in transport and other responses?
- Are there strongly interacting topological phases in real materials?

# Ask Professor Joe

SAMR in CeAlGe, a  
magnetic Weyl material

What's going on??

- A) Device is sensing magnetic poles
- B) It's the chiral anomaly
- C) Protection from backscattering leads to low resistance away from high symmetry directions
- D) The field is sweeping across Ising ordered "nematic"-like phases resulting in domain wall resistance
- E) Joe's cell phone interfered with the signal from the cryostat



# Themes of modern QMs

- Order
- Topology
- Entanglement
- Correlations

# Entanglement

$$\text{EPR} \quad |\Psi\rangle = \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle)$$

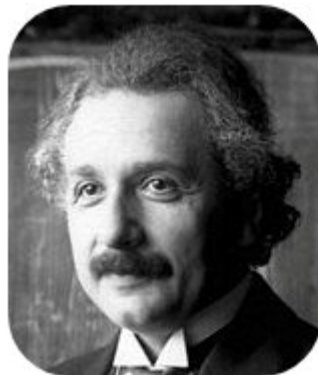
entangled: cannot be written as  
a product state

# Entanglement

EPR  $|\Psi\rangle = \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle)$



??where is the information??



A. Einstein



B. Podolsky



N. Rosen

# Many Body Entanglement



Phil Anderson, 1973

a “quantum liquid” of spins

$$\text{blue oval} = \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle)$$

$$\Psi = \text{diagram 1} + \text{diagram 2} + \dots$$

The diagram shows two lattice configurations of blue ovals (representing singlet states) on a triangular lattice. In the first configuration, the ovals are arranged in a regular pattern. In the second configuration, the ovals are shifted, representing a different state in the superposition.

**Resonating Valence Bond** state

# Many Body Entanglement



Phil Anderson, 1973

a “quantum liquid” of spins

$$\text{blue oval} = \frac{1}{\sqrt{2}} (|\uparrow\downarrow\rangle - |\downarrow\uparrow\rangle)$$

$$\Psi = \text{[diagram 1]} + \text{[diagram 2]} + \dots$$

The diagram shows the wavefunction  $\Psi$  as a sum of states. Each state is represented by a triangular lattice of blue ovals, which represent singlet spin pairs. In the first diagram, the ovals are arranged in a regular, periodic pattern. In the second diagram, the ovals are shifted or rearranged, representing a different configuration of the many-body state. The ellipsis indicates that there are many more such states in the sum.

**Resonating Valence Bond** state



# Ordinary (local) Matter

We can consistently assign local properties (elastic moduli, etc.) and obtain all large-scale properties



- Measurements far away do not affect one another
- From local measurements we can deduce the global state

# Ordinary (local) Matter

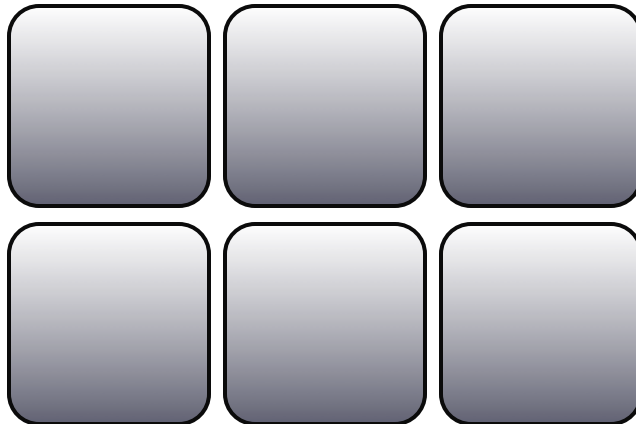
Hamiltonian is local

$$H = \sum_{\mathbf{x}} \mathcal{H}(x) \quad \mathcal{H}(x) \text{ has local support near } x$$

Ground state is "essentially"  
a product state

$$|\Psi\rangle = \otimes_A |\psi\rangle_A$$

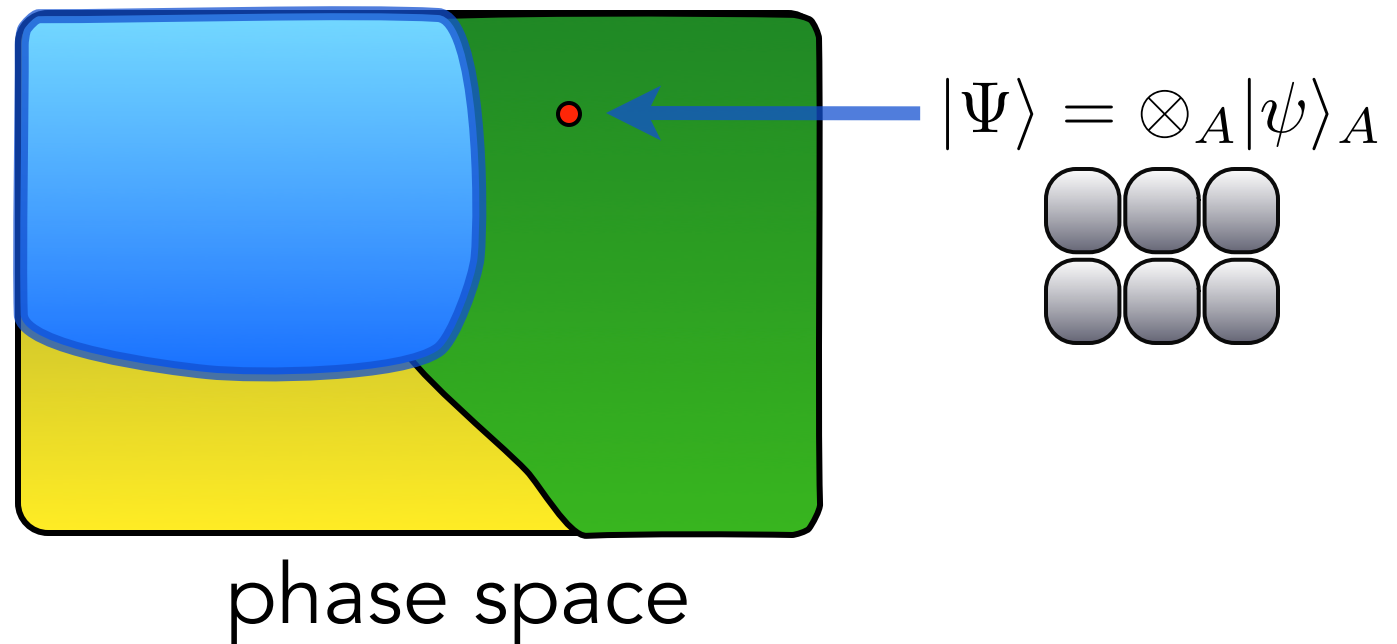
most insulators  
are like this



no entanglement  
between blocks

# "Essentially" a product state?

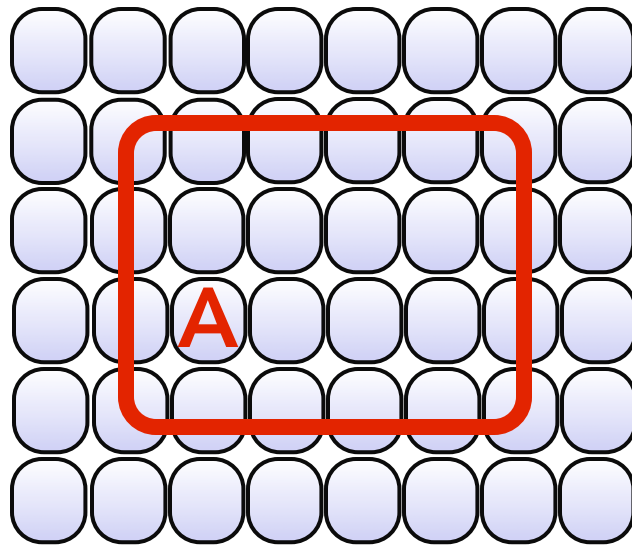
- Adiabatic continuity



n.b. This is not true for gapless fermi systems

# “Essentially” a product state?

- Entanglement scaling



$$\rho_A = \text{Tr}_{\bar{A}} |\Psi\rangle\langle\Psi|$$

$$S(A) = -\text{Tr}_A (\rho_A \ln \rho_A)$$

$$S(A) \sim \sigma L^{d-1} \quad \text{area law}$$

satisfied with exponentially small corrections

# Best example: ordered magnet

Hamiltonian

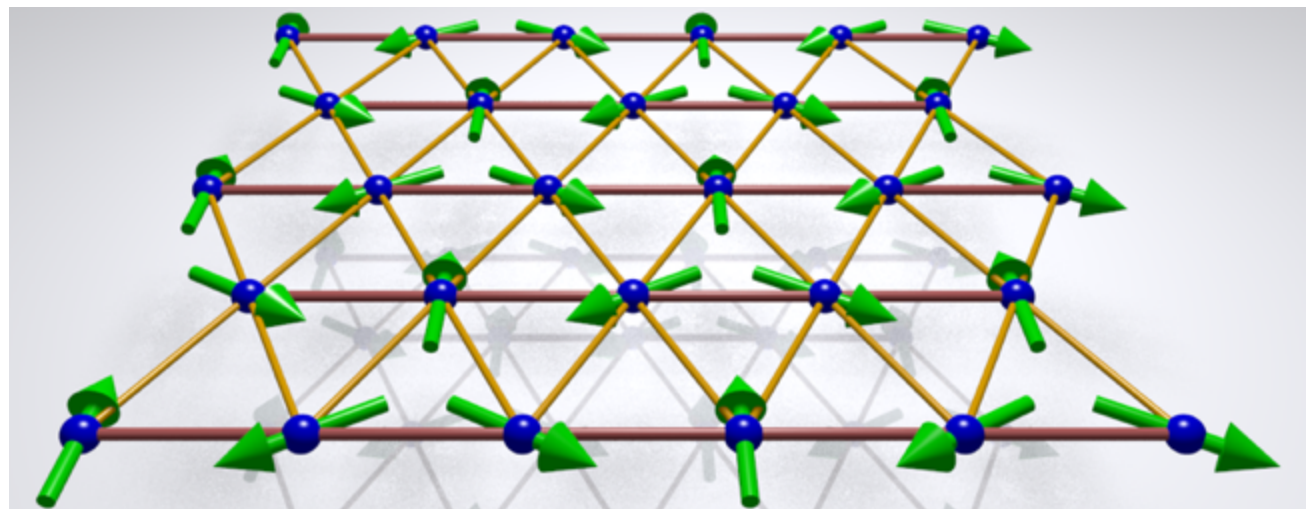
$$H = \sum_{(ij)} J_{ij} \mathbf{S}_i \cdot \mathbf{S}_j$$

exchange is short-  
range: local

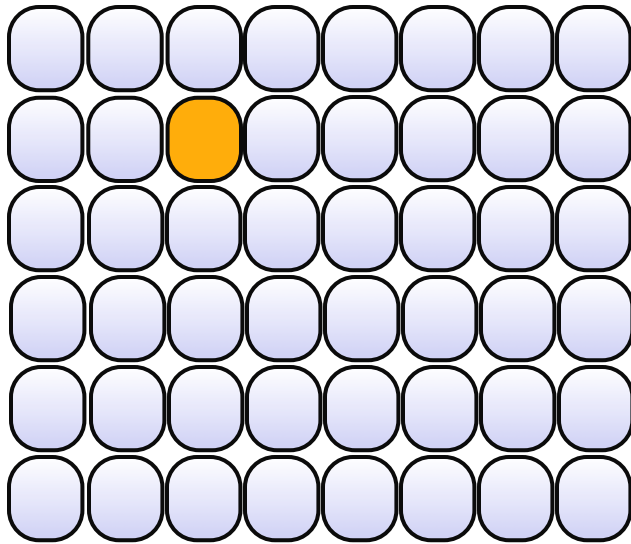
ordered state

$$|\Psi\rangle \approx \bigotimes_i |\mathbf{S}_i \cdot \hat{n}_i = +S\rangle$$

block is a single  
spin



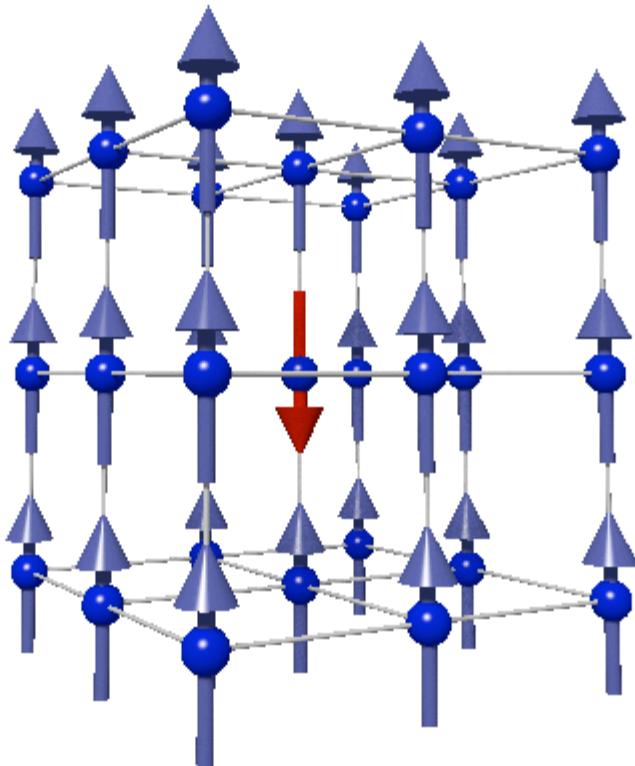
# Quasiparticles



excited states  $\sim$  excited levels of one block

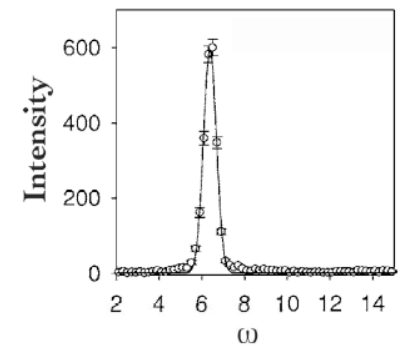
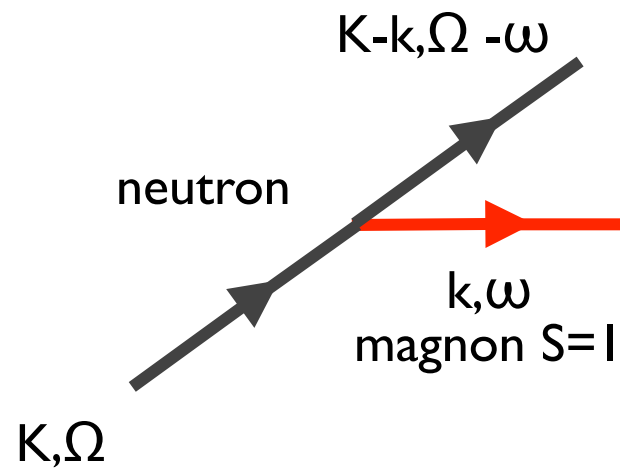
- local excitation can be created with operators in one block
- localized excitation has discrete spectrum with non-zero gap, and plane wave forms sharp band
- quantum numbers consistent with finite system: no emergent or fractional quantum numbers

# Spin wave



$$\omega(k) \approx \Delta - 2t \cos k_x a - \dots$$

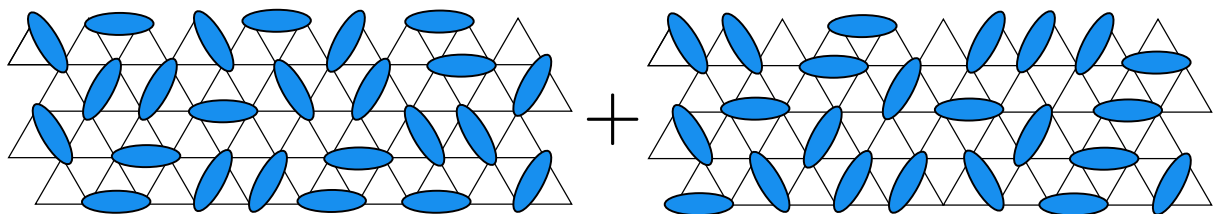
$$|f\rangle = S_k^+ |i\rangle$$



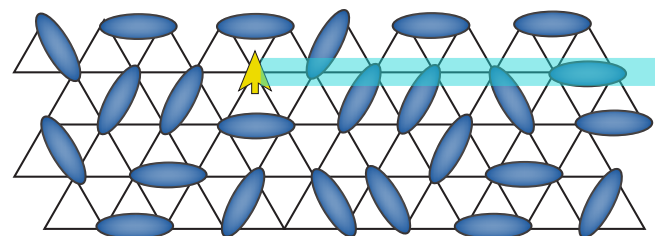
Line shape in Rb<sub>2</sub>MnF<sub>4</sub>

# Quantum spin liquid

Entanglement  $\rightarrow$  non-local excitation

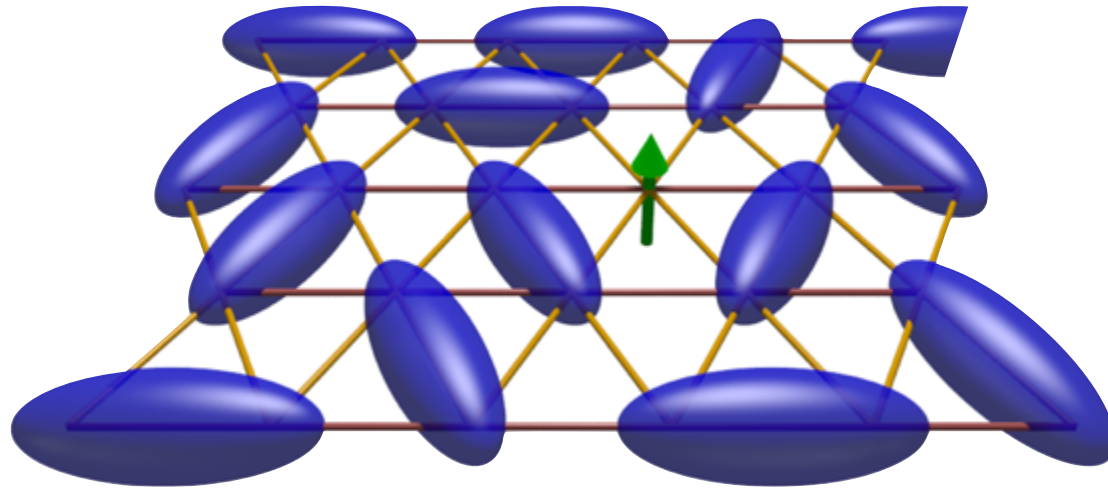
$$\Psi = \text{[Diagram 1]} + \text{[Diagram 2]} + \dots$$




$$\Psi = \text{[Diagram 3]} + \dots \quad \text{"spinon"}$$


"quasiparticle" above a non-zero gap

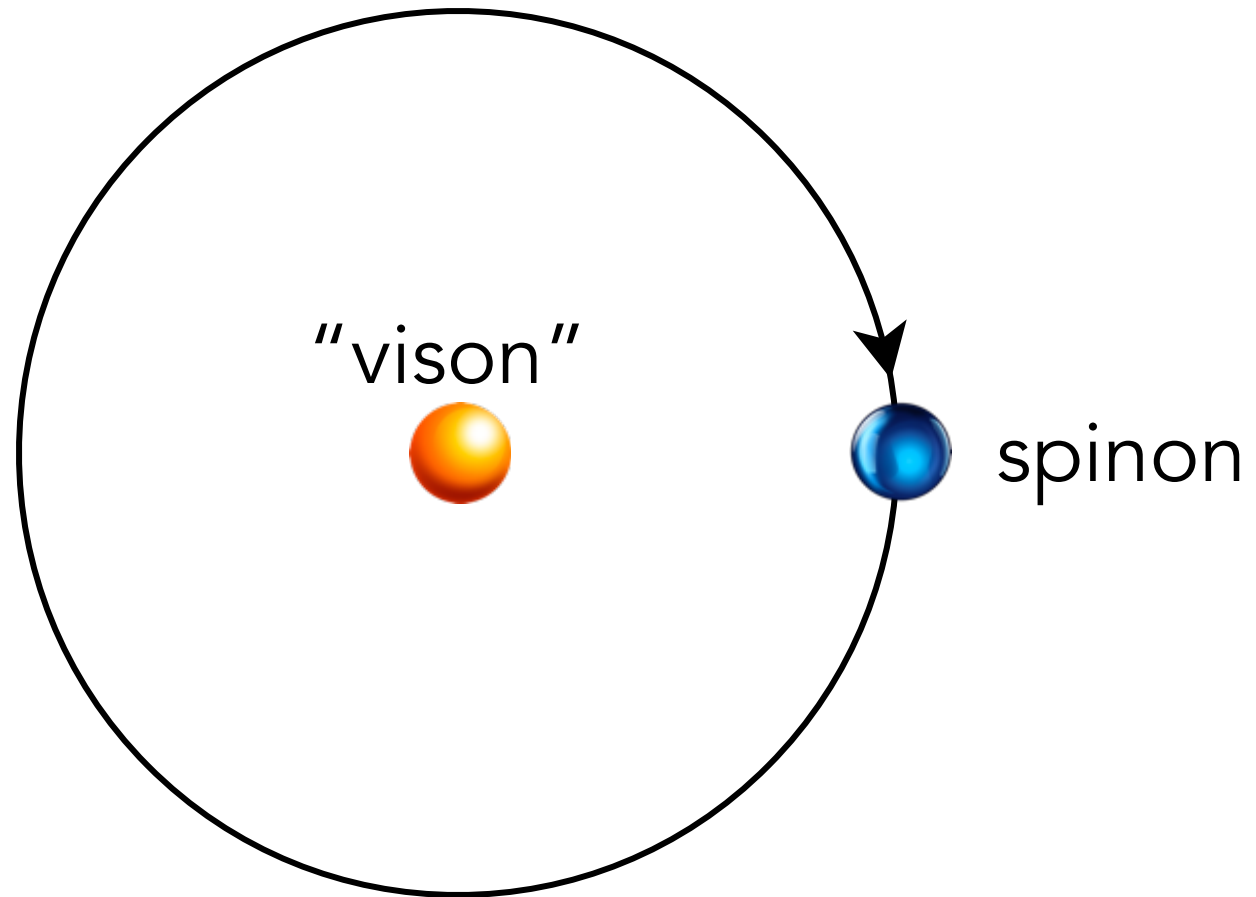
# Fractional quantum number



excitation with  $\Delta S = 1/2$   
not possible for any finite  
cluster of spins  
always created in pairs by any  
local operator

# Anyons

A characteristic of  
"intrinsic  
Topological Order"



$$\Psi \rightarrow -\Psi$$

"mutual semions"

# Where does this name vison come from?

- A) Because they occur at high pressures like in a vise
- B) Vision was taken by the Marvel character
- C) Vortex+Ising = vison
- D) Named after the zoological name for the mink, because their long-range statistical interaction extends like the famous fur
- E) I don't know but why do people keep naming particles that don't exist?



X.-G. Wen

(intrinsic)

# Topological phases



A. Kitaev



Warning: this is a different meaning of topological!

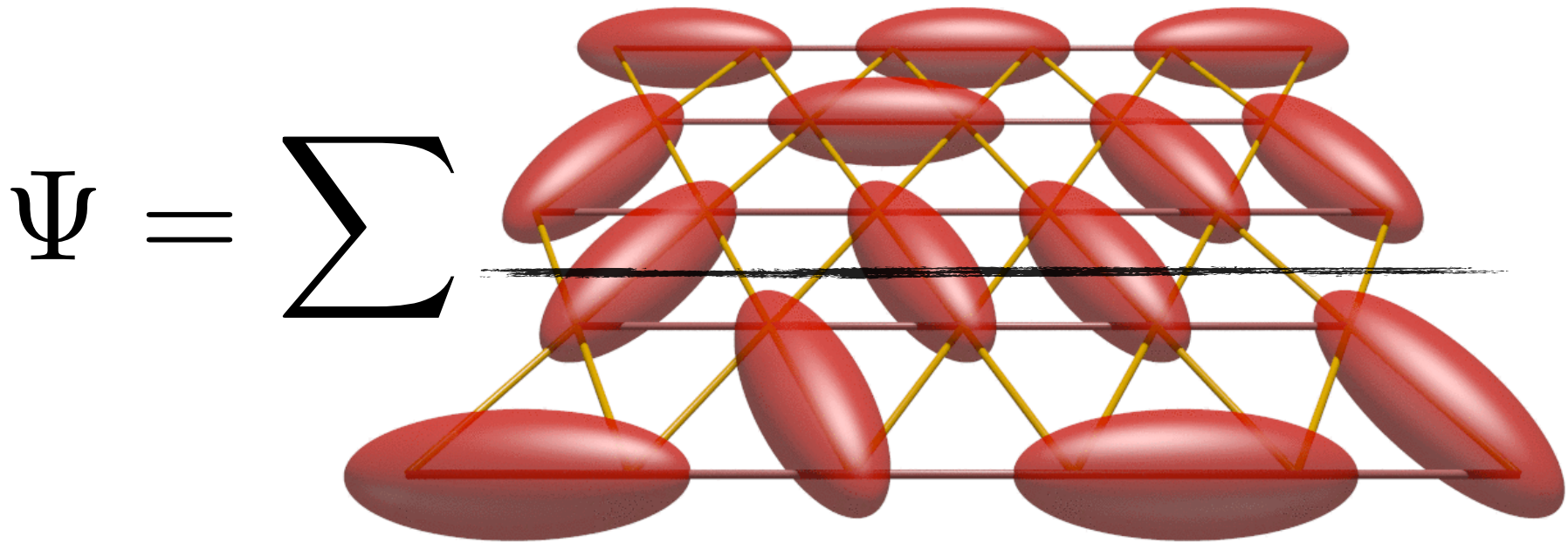
Anderson's RVB state is thus an example of a "topological phase" - the best understood sort of QSL

Understood and classified by anyons and their braiding rules in 2d

$$\begin{array}{c} e \quad m \\ \diagdown \quad \diagup \\ \diagup \quad \diagdown \\ e \quad m \end{array} = - \begin{array}{cc} e & m \\ | & | \\ e & m \end{array}$$

$$\begin{array}{cc} e m & e m \\ \diagdown & \diagup \\ \diagup & \diagdown \\ e m & e m \end{array} = \begin{array}{cc} e m & e m \\ | & | \\ \diagdown & \diagup \\ e m & e m \end{array} = - \begin{array}{ccc} e m & e m & e m \\ | & | & | \\ e m & e m & e m \end{array}$$

# Stability

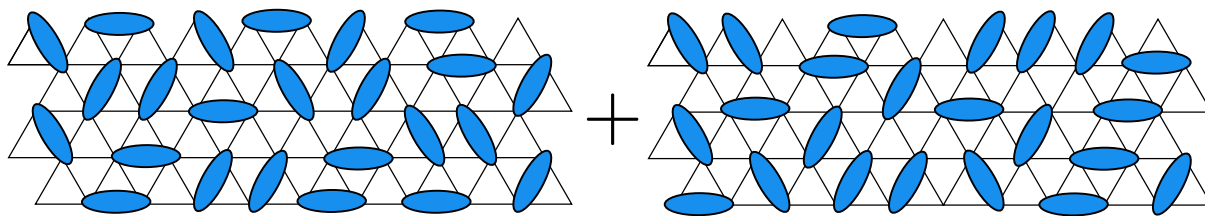


Robustness arises from topology: a QSL is a stable *phase* of matter (at  $T=0$ )

How stable is a topological QSL? Which of the choices below will make a 2d QSL become the same as a paramagnet?

- A) Apply some weak strain
- B) Apply a small magnetic field
- C) Add some weak randomness to the bonds
- D) Beat the heck out of it with a hammer
- E) Warm it up

# Quantum spin liquid

$$\Psi = \text{[Diagram 1]} + \text{[Diagram 2]} + \dots$$


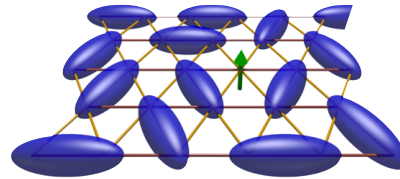
The diagram illustrates the wavefunction  $\Psi$  of a quantum spin liquid as a superposition of different spin configurations. It shows two triangular lattices, each with blue ellipses representing spins. The first lattice shows a specific arrangement of spins, and the second lattice shows a different arrangement. The two lattices are added together, followed by an ellipsis indicating that there are many more such configurations in the superposition.

For  $\sim 500$  spins, there are more amplitudes than there are atoms in the visible universe!

Different choices of amplitudes can realize different QSL phases of matter.

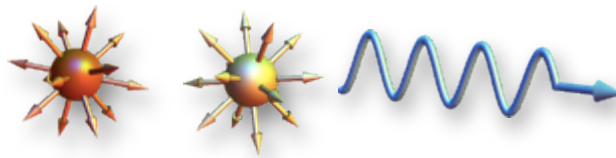
# Varieties of QSLs

- Topological QSLs



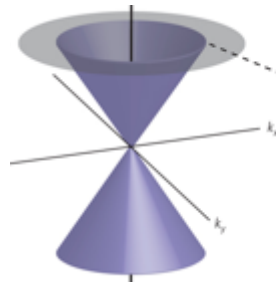
anyonic  
spinons

- U(1) QSL



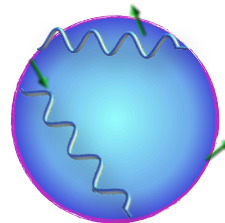
electric+magnetic  
monopoles, photon

- Dirac QSLs



strongly  
interacting  
Dirac fermions

- Spinon Fermi surface

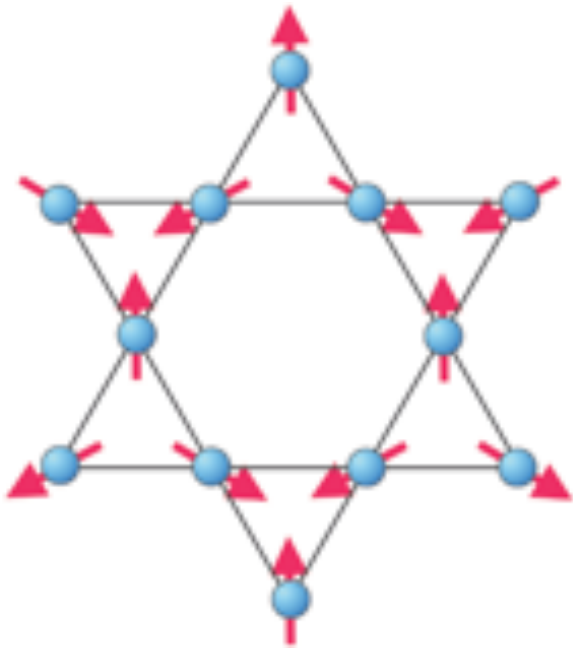


non-Fermi  
liquid "spin  
metal"

# QSL experiments

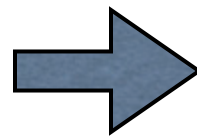
- This is a difficult subject, if you want a challenge!
- Discuss three examples:
  - Kagomé lattice herbertsmithite
  - Organic triangular lattice
  - $\alpha$ - $\text{RuCl}_3$  Kitaev magnet

# Kagomé antiferromagnet



$$H = J \sum_{\langle ij \rangle} \mathbf{S}_i \cdot \mathbf{S}_j + \dots$$

Very large classical  
degeneracy

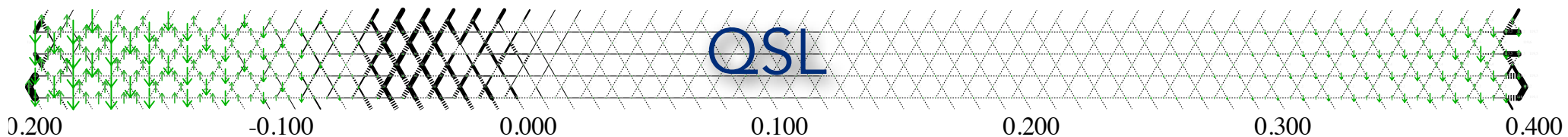


likely to be a QSL

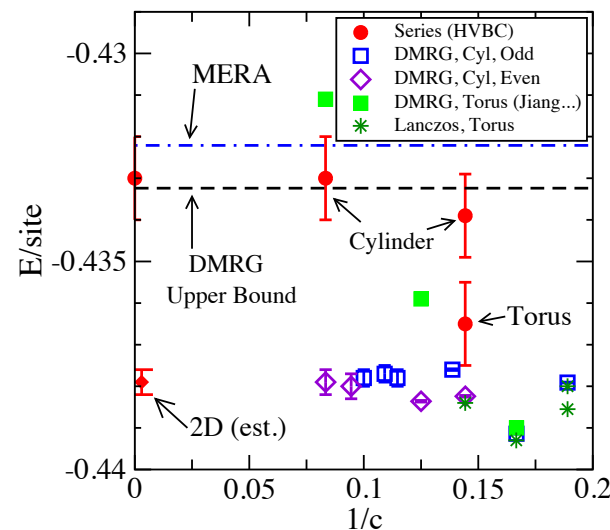
V. Elser, 1989 + many many others

# $S=1/2$ kagomé AF

- Rather definitive evidence for QSL by DMRG



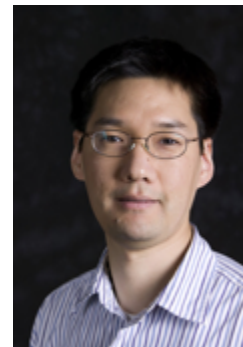
© Steve White



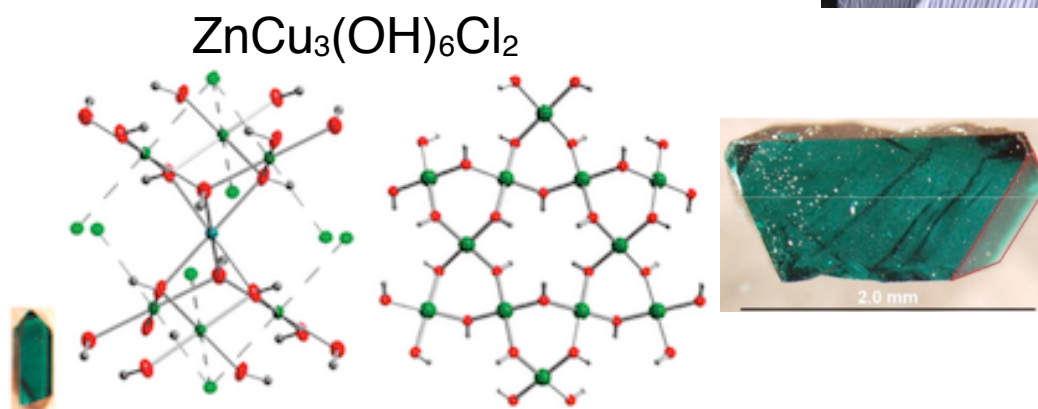
*S. Yan et al, 2010*

many other studies support  
existence of some QSL phase

# Herbertsmithite

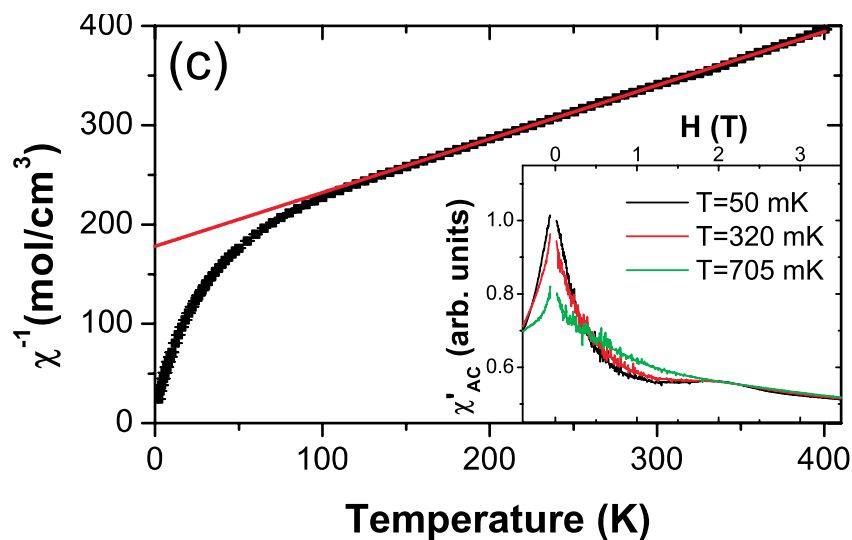


kagomé layers of Cu  
 $S=1/2$  spins, separated  
by non-magnetic Zn



Heisenberg-like  
with  $J \sim 200\text{K}$

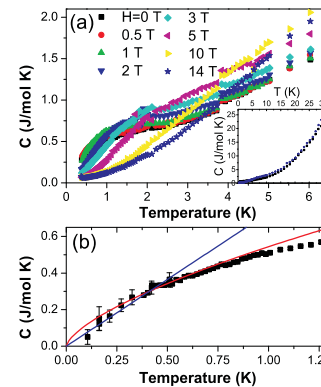
no order down to  
50mK



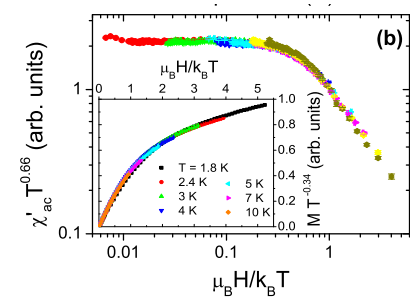
Helton et al, 2007

# Herbertsmithite

Lots of early evidence  
for gaplessness



Helton et al, 2007

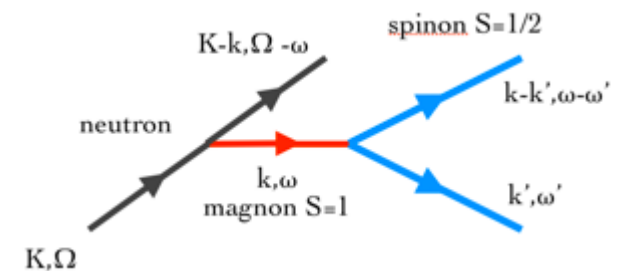
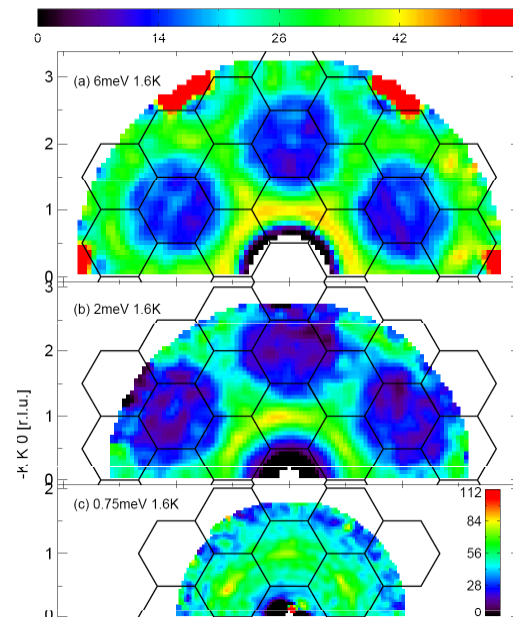


Helton et al, 2010

Single crystal INS

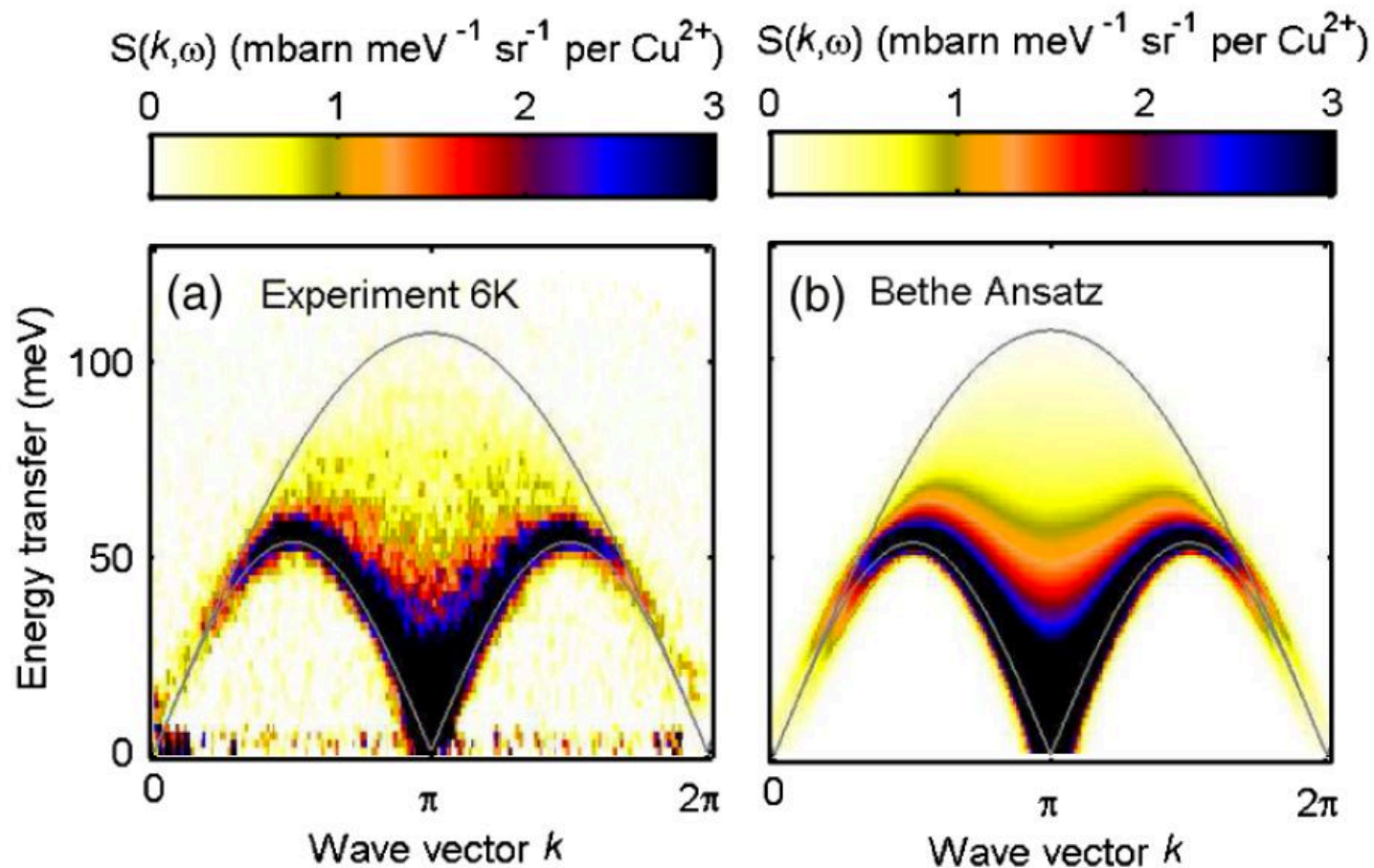
smooth continuum  
scattering

T-H Han et al, 2012



continuum scattering  
expected  
...but probably with more  
structure?

# For comparison: 1d spinons

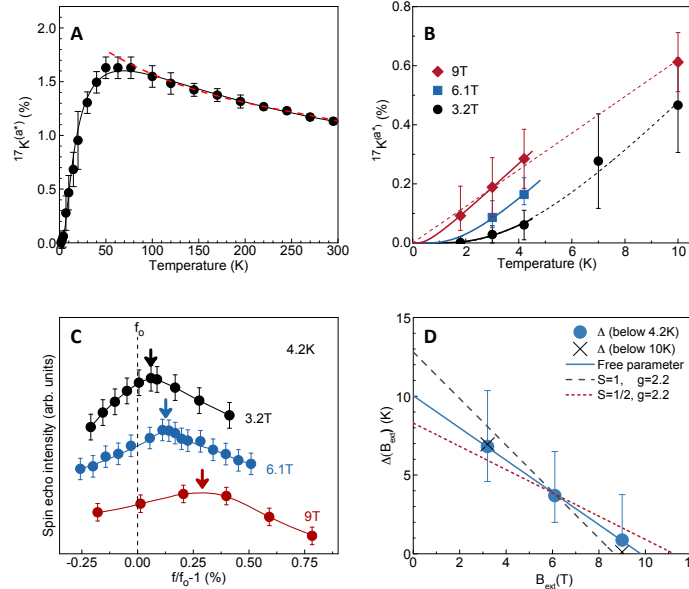


$\text{KCuF}_3$  - B. Lake *et al*, 2013

# Herbertsmithite

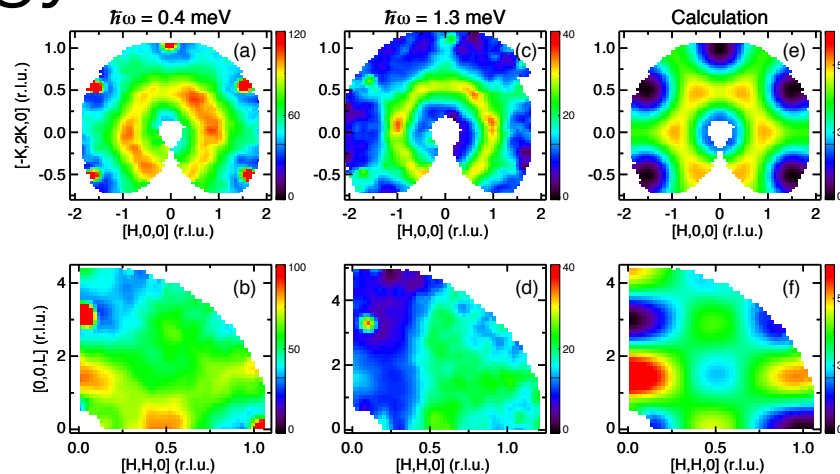
## Single crystal NMR

M. Fu *et al*, 2015



estimate gap  $\sim$   
10K

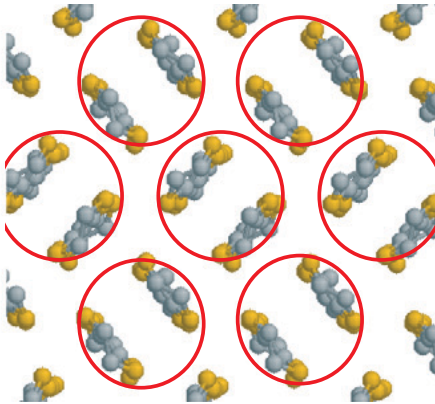
## Low energy INS



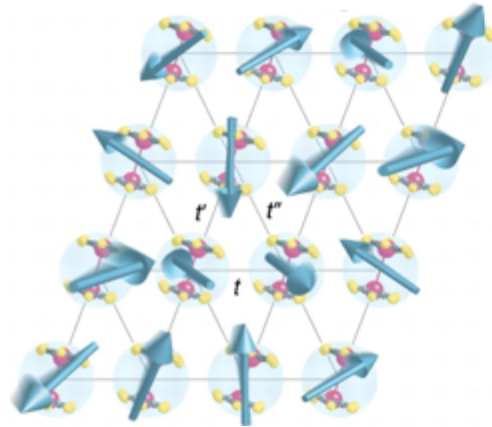
claim to separate  
impurity signal  
below 0.7meV

T-H Han *et al*, 2015

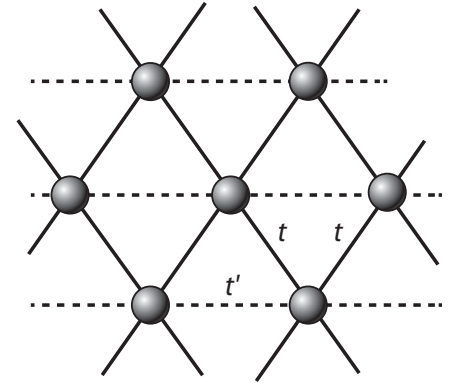
# Organics



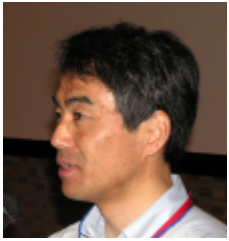
$\kappa\text{-(ET)}_2\text{X}$



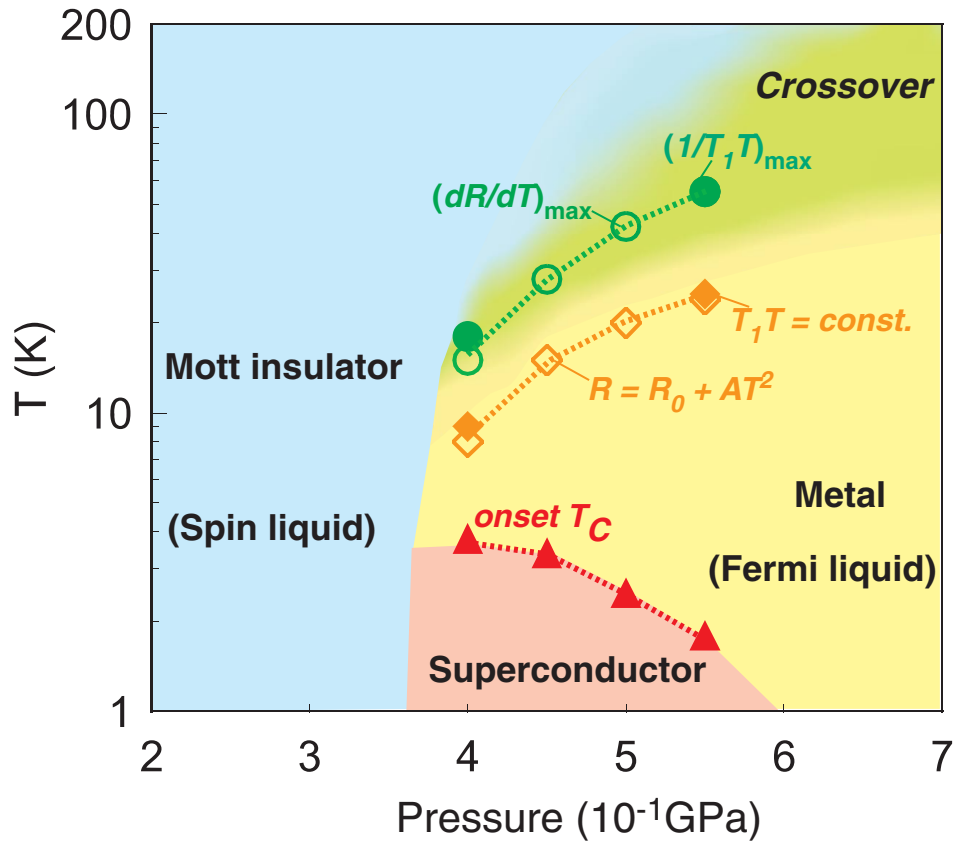
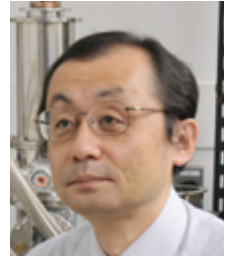
$\beta'\text{-Pd(dmit)}_2$



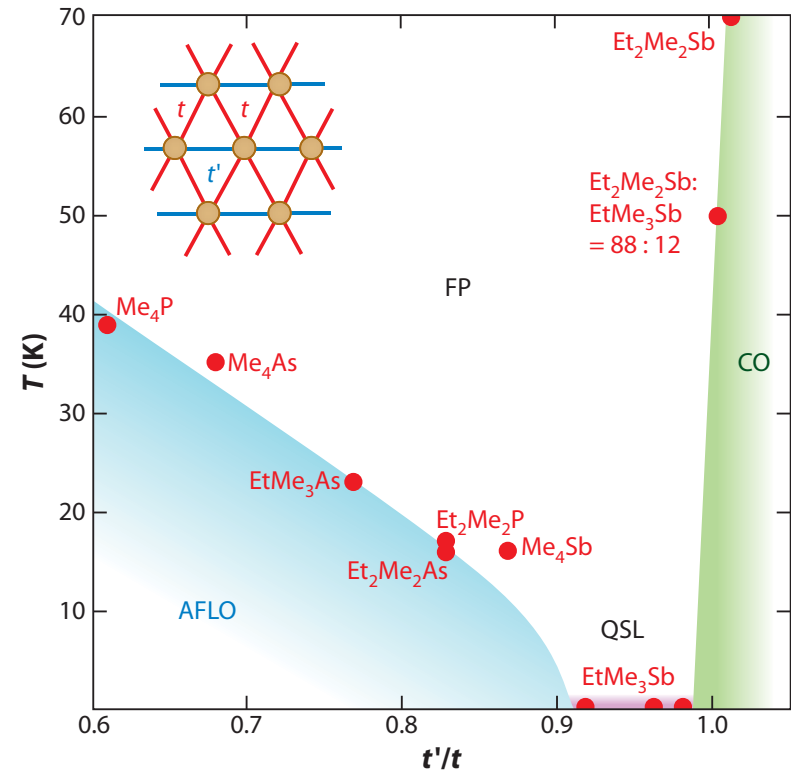
- Molecular materials which behave as effective triangular lattice  $S=1/2$  antiferromagnets with  $J \sim 250\text{K}$
- significant charge fluctuations



# Organics

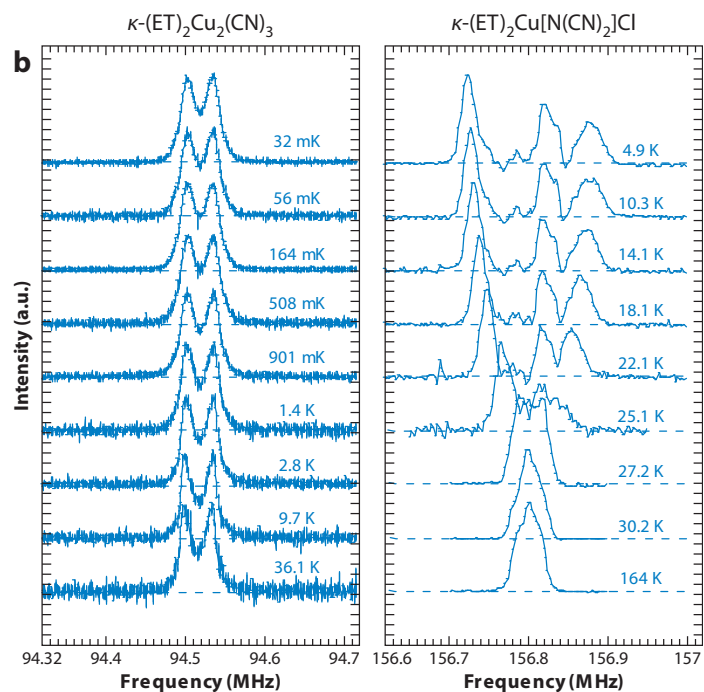


K. Kanoda group (2003-)

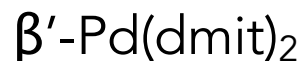
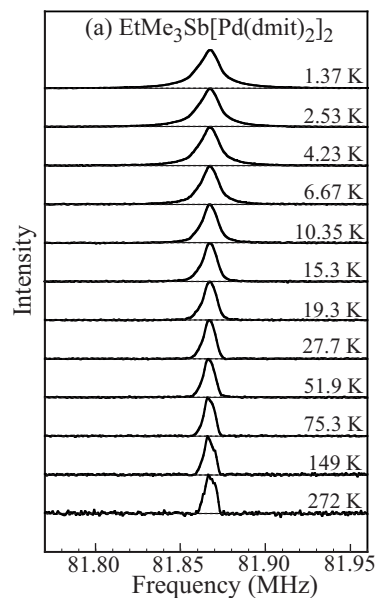


R. Kato group (2008-)

# NMR lineshapes

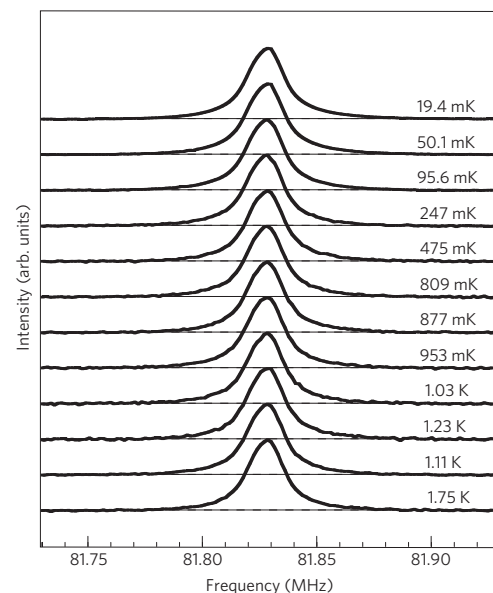


Y. Shimizu  
*et al*, 2003  $^1\text{H}$  NMR



T. Itou *et al*,

2008, 2010

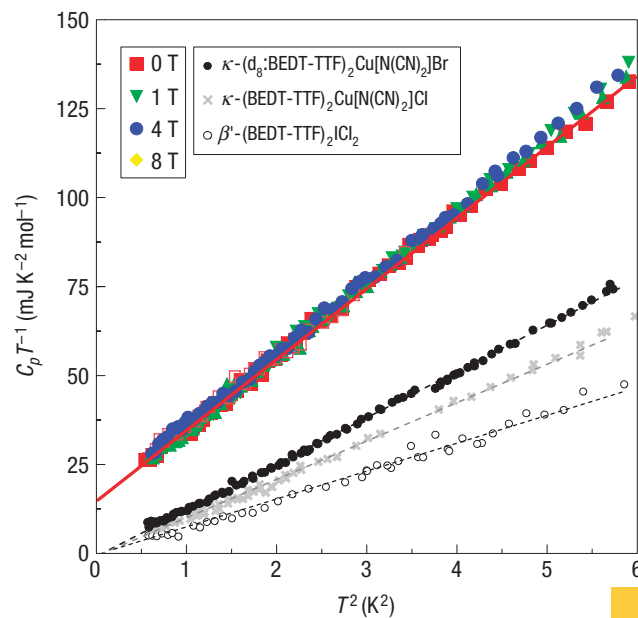


$^{13}\text{Cs}$  NMR

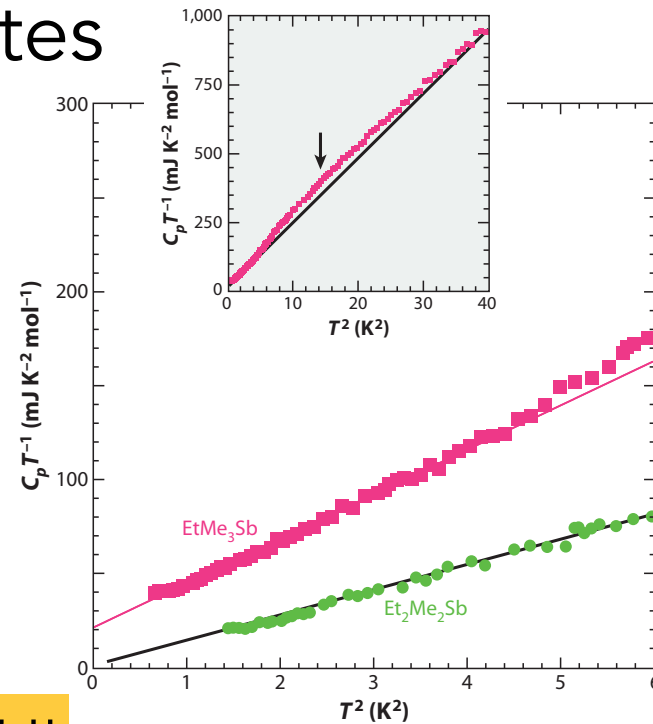
Evidence for lack of static moments:  $f > 1000$ !

# Specific Heat

- $C \sim \gamma T$  indicates gapless behavior with large density of states



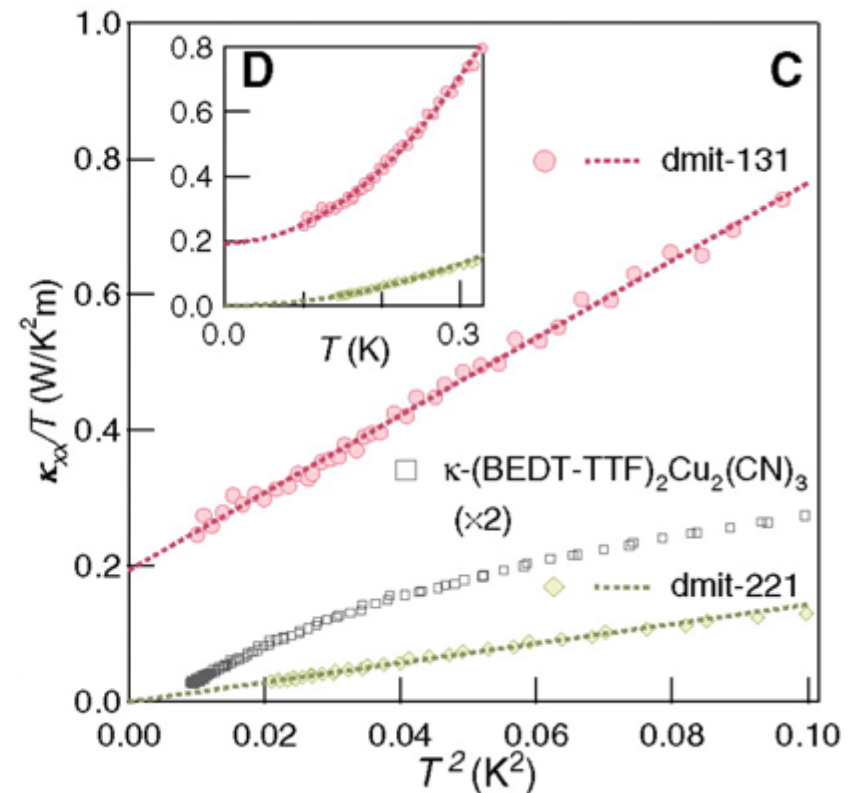
$\gamma_{Cu} \sim 0.7$  !!



S. Yamashita et al, 2008

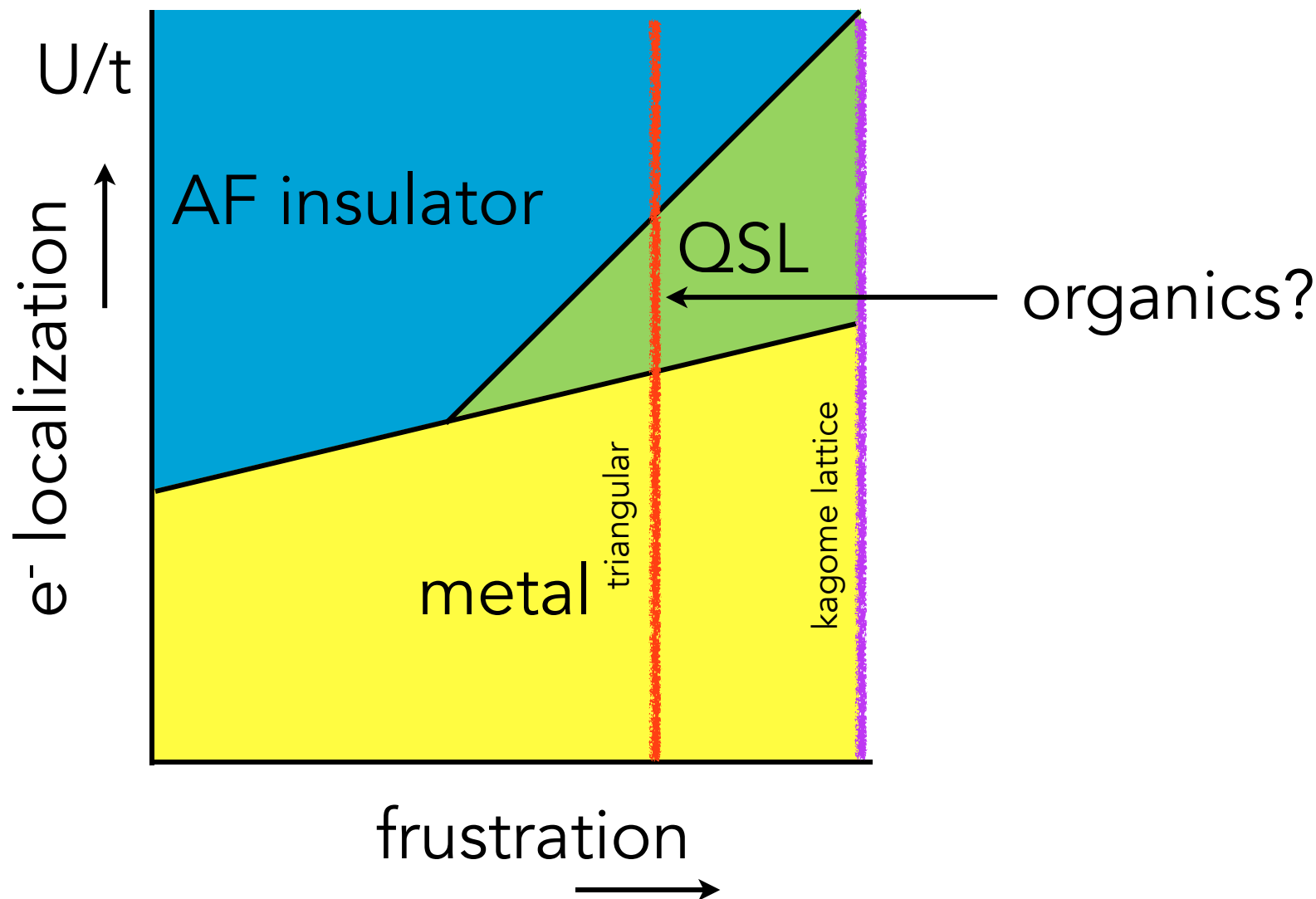
# Thermal conductivity

- Huge linear thermal conductivity indicates the gapless excitations are propagating, at least in dmit
- Estimate for a *metal* would correspond to a mean free path  $l \sim 1 \mu\text{m} \approx 1000 a$  !

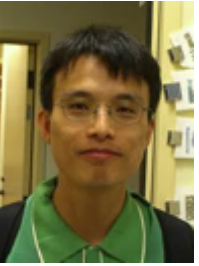


M. Yamashita et al, 2010

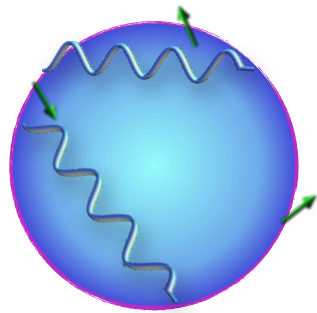
# Charge fluctuations



# Organics - Theory



- RVB/QSL state:
  - Motrunich, Lee+Lee: (2005) "uniform RVB"
  - It is described by a "**Fermi sea**" of **spinons** coupled to a U(1) gauge field
- The anomalous thermal conductivity may be a window into an emergent fermi surface in an insulator!



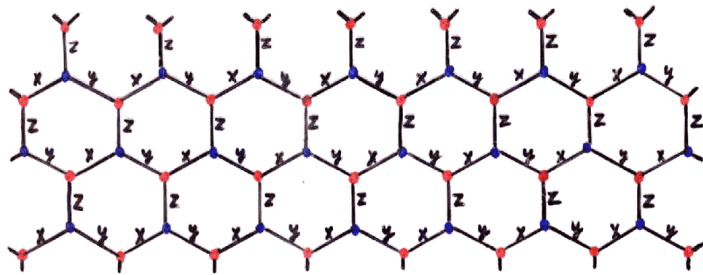


# Kitaev model

Kitaev's honeycomb model

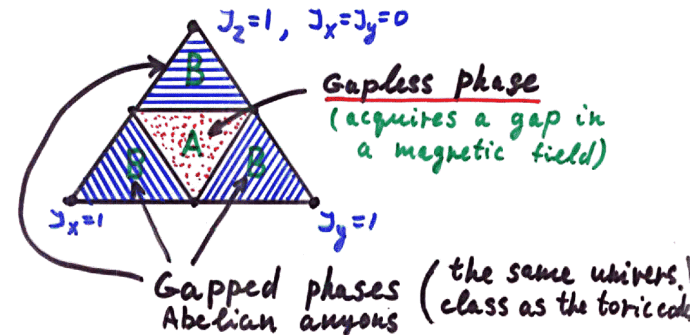
$$H = \sum_{i,\mu} K_{\mu} \sigma_i^{\mu} \sigma_{i+\mu}^{\mu}$$

## 1. The model



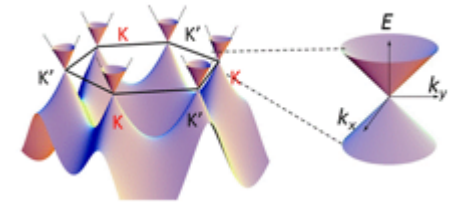
Spin  $\frac{1}{2}$  on each site.

## Phase diagram



exact parton construction  $\sigma_i^{\mu} = i c_i c_i^{\mu}$   $c_i c_i^x c_i^y c_i^z = 1$

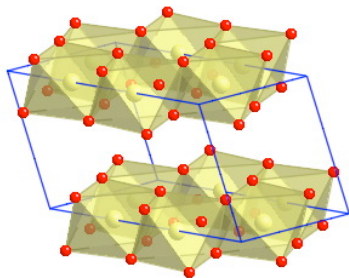
physical Majoranas  $H_m = K \sum_{\langle ij \rangle} i c_i c_j$



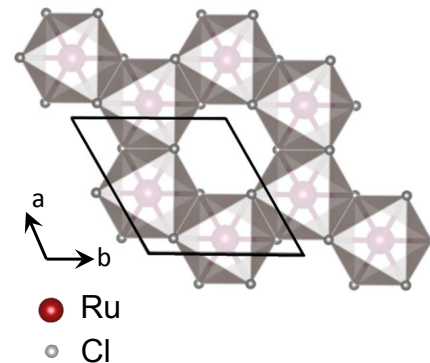
# Kitaev Materials

Jackeli, Khaliullin  
2009

Showed that Kitaev interaction can be large in edge-sharing octahedra with large spin-orbit-coupling

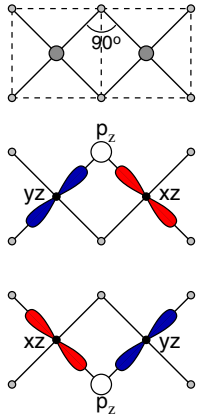


$\text{Na}_2\text{IrO}_3$ ,  
 $(\alpha, \beta, \gamma)$ -  
 $\text{Li}_2\text{IrO}_3$



$\alpha\text{-RuCl}_3$

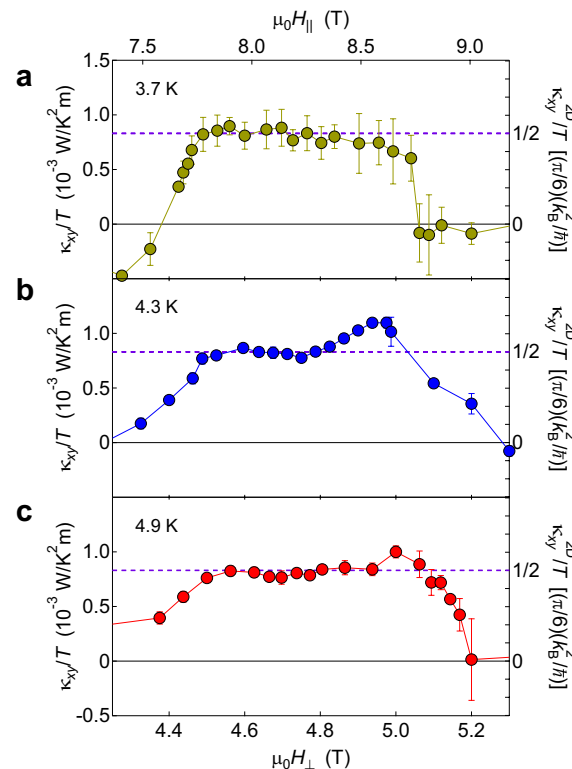
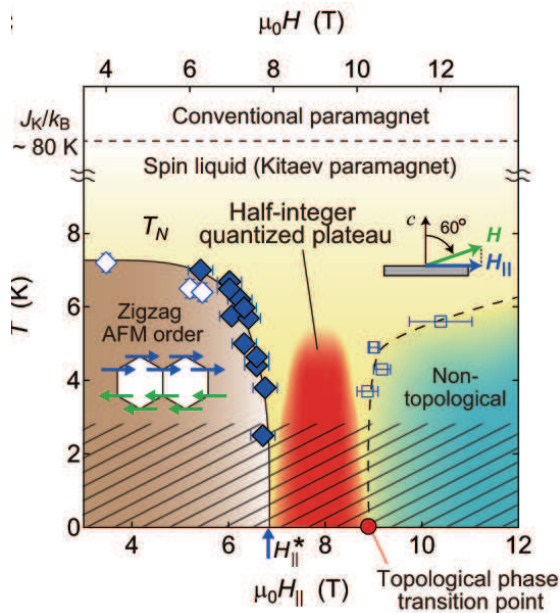
Y.-J. Kim...



Honeycomb and hyper-honeycomb structures

# Thermal Hall

- Very recent experiment on  $\alpha$ -RuCl<sub>3</sub>



$$\frac{\kappa_{xy}}{T} = c \frac{\pi}{6} \frac{k_B^2}{\hbar}$$

$c=1/2$  is expected for a *chiral Majorana fermion edge*, characteristic of "Ising anyons"

# Entanglement Issues

- Can you definitively identify highly entangled phases? How?
- Can you measure entanglement?
- What are the links between entanglement, thermalization, and hydrodynamics?

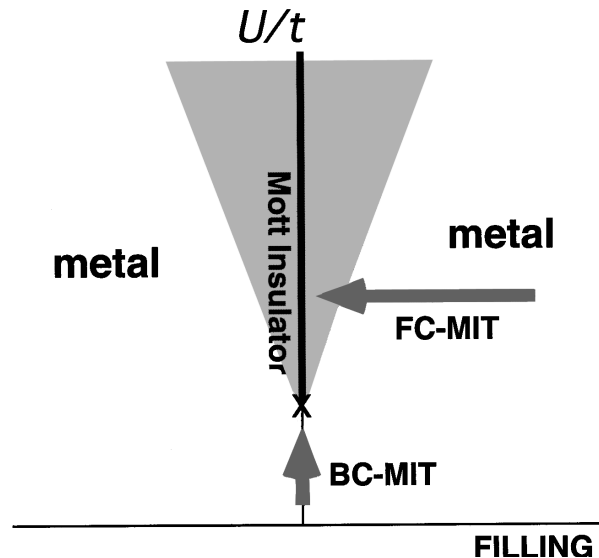
# Correlation

- Many phenomena result from interactions that are neither topological nor simply described in terms of entanglement (which can be thought of as a particular type of correlation)
  - Mott metal-insulator transition
  - Heavy electrons and non-Fermi liquids
  - Fluctuating orders and pseudogaps
  - Low dimensional systems

# Mott Transition

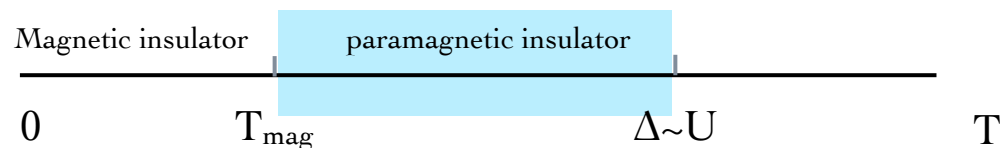
- Hubbard model

$$H = -t \sum_{\langle ij \rangle} c_{i\alpha}^\dagger c_{j\alpha} + U \sum_i n_{i\uparrow} n_{i\downarrow}$$



Imada, Fujimori, Tokura 1998

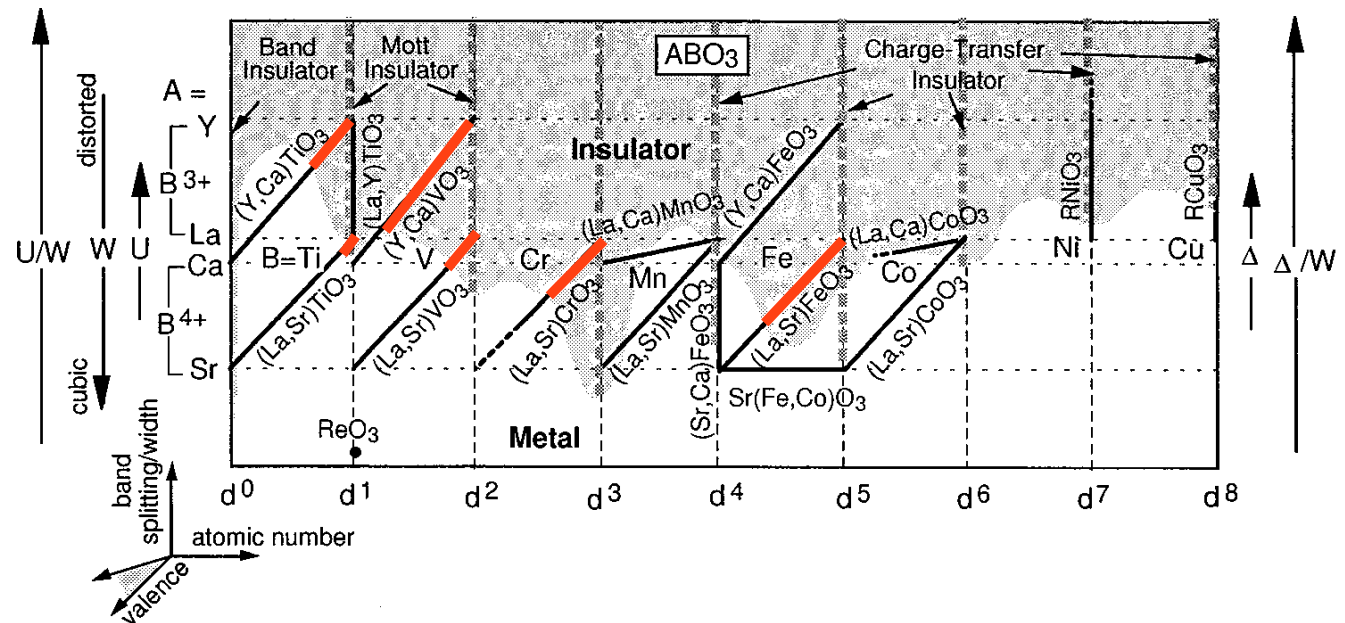
- ✱ Physically, a Mott Insulator is one which is insulating due to interaction-induced localization, not due to band physics
- ✱ This is a question of energy scales
- ✱ Deep in the Mott state:



# Mott Transition

- And old subject

e.g.  
perovskites

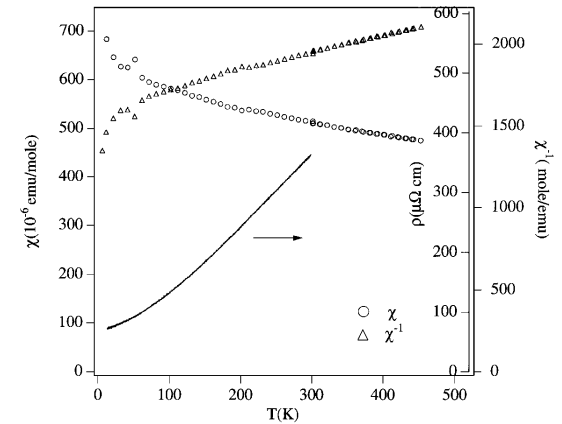
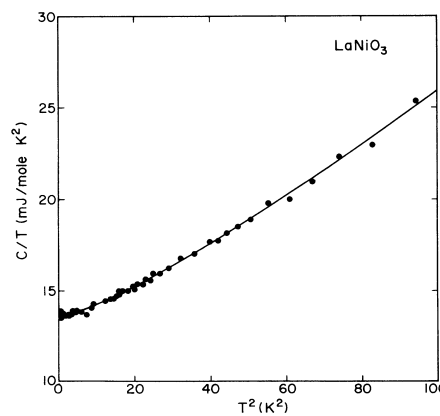


- But still a key theme, and now more accessible in new materials, via new approaches

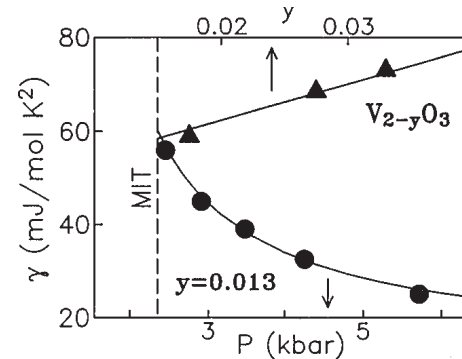
# Heavy Electrons

- Mass enhancement near Mott transition

LaNiO<sub>3</sub>:  
 $\gamma/\gamma_{\text{band}} = 10$



V<sub>2</sub>O<sub>3</sub>



# Heavy Fermions

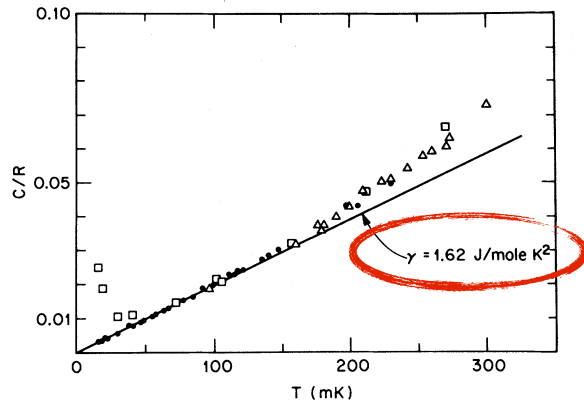


FIG. 1. Specific heat of  $\text{CeAl}_3$  at very low temperatures in zero field ( $\bullet, \Delta$ ) and in 10 kOe ( $\square$ ).

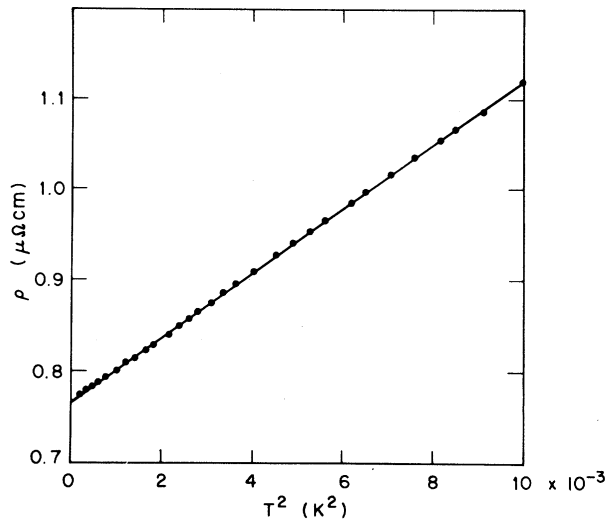


FIG. 3. Electrical resistivity of  $\text{CeAl}_3$  below 100 mK, plotted against  $T^2$ .

$$C \sim \gamma T$$

$$\rho(T) - \rho(0) \sim AT^2$$

Both  $\gamma$  and  $A$  **huge**

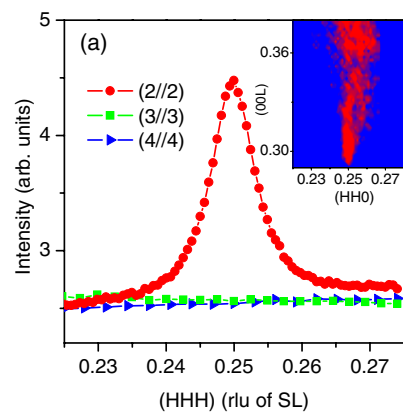
Behave like Fermi liquid with tiny  $E_F$  and large electron mass, but only for  $T \ll E_F$

Common in "Kondo lattice" materials

# Oxide heterostructures

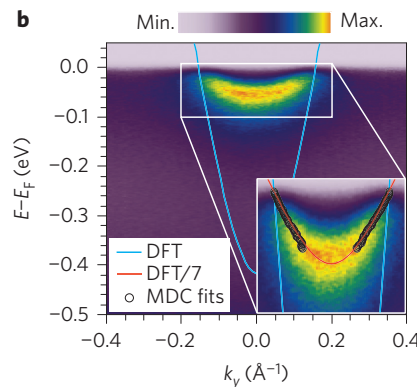
- New handle on Mott materials from oxide MBE

example:  $\text{LaNiO}_3$  - just on the metallic side of Mott transition in bulk

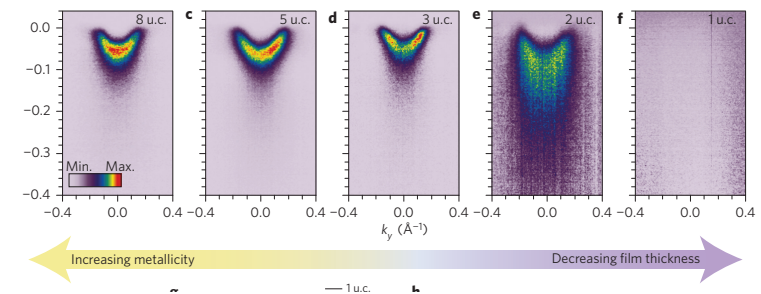


AF order for films  
<4uc thickness.

Frano *et al*, 2013



Large mass in 8uc  
thick films

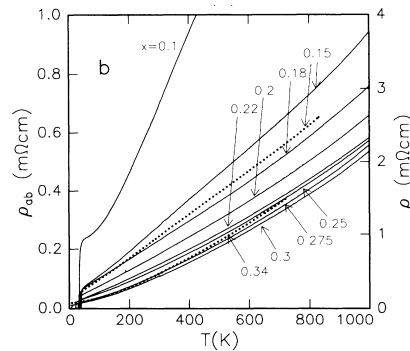


Metal-insulator  
transition at 2uc  
thickness

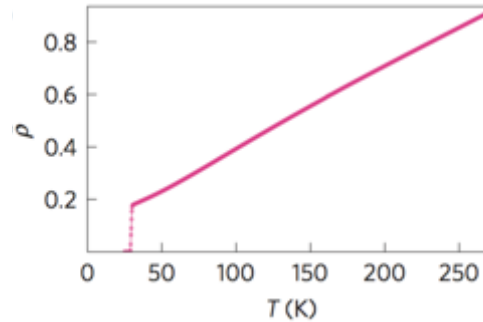
King *et al*, 2014 [Schlom, Shen]

# Non-Fermi Liquids

“strange metal”



LSCO Takagi et al, 1992



$\text{BaFe}_2(\text{As}_{1-x}\text{P}_x)_2$ , Hayes et al, 2016

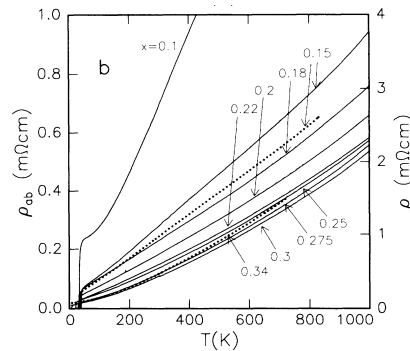
$$\frac{1}{\tau} \sim T ?$$

T-linear resistivity/scattering rate:

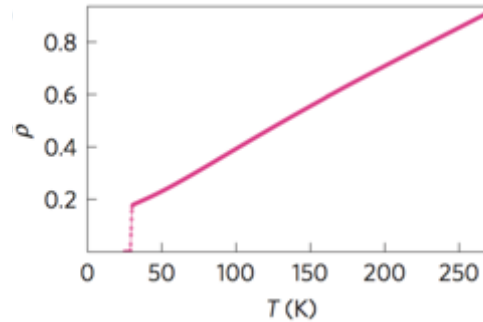
- Many materials
- Often nearby to unconventional superconductivity
- Symptom of a different type of metal? Or of a quantum critical point?

# Non-Fermi Liquids

"strange metal"



LSCO Takagi et al, 1992



BaFe<sub>2</sub>(As<sub>1-x</sub>P<sub>x</sub>)<sub>2</sub>, Hayes et al, 2016

$$\frac{1}{\tau} \sim T ?$$

## Quasiparticles?

Green's function

$$G_R(\mathbf{k}, \omega) = \frac{1}{\omega - \epsilon_{\mathbf{k}} - \Sigma(\mathbf{k}, \omega) - i\delta}$$

self-energy

Spectral density

$$\rho(\mathbf{k}, \omega) = \frac{-2\Sigma''(\mathbf{k}, \omega)}{(\omega - \epsilon_{\mathbf{k}} - \Sigma'(\mathbf{k}, \omega))^2 + (\Sigma''(\mathbf{k}, \omega))^2}$$

$$\Sigma''_{\text{FL}} \sim \omega^2 \sim T^2$$

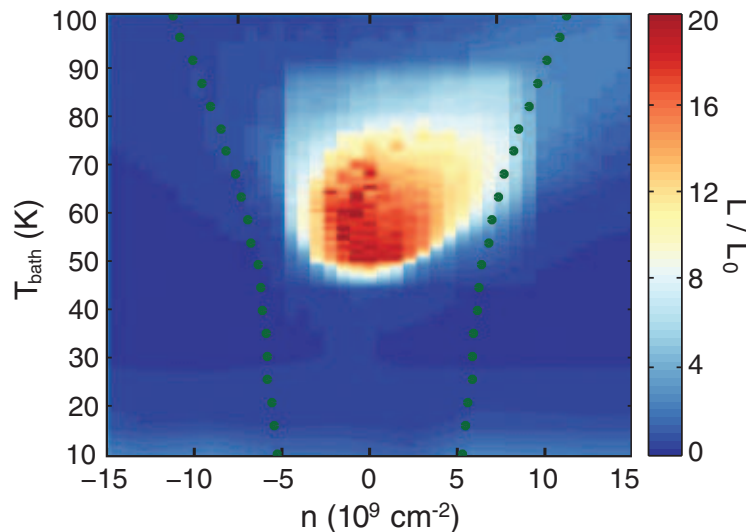
$$\text{here } \Sigma''_{\text{FL}} \sim T$$

# What is the mechanism of the strange metal?

- A) Electron nematic quantum criticality
- B) It's dual to a black hole in anti-de-Sitter space
- C) Dynamical mean field theory
- D) Interaction with an emergent gauge field
- E) Measurement error

# Hydrodynamic flow

- Low density, ultra-clean fluids - here graphene: e-e scattering creates *viscous flow*



Red region: thermal conductivity much larger than electrical conductivity. Indicates e-e fluid regime

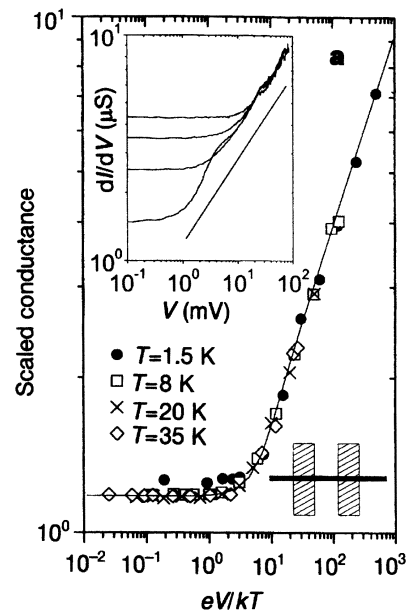
nple.

Crossno *et al*, 2016

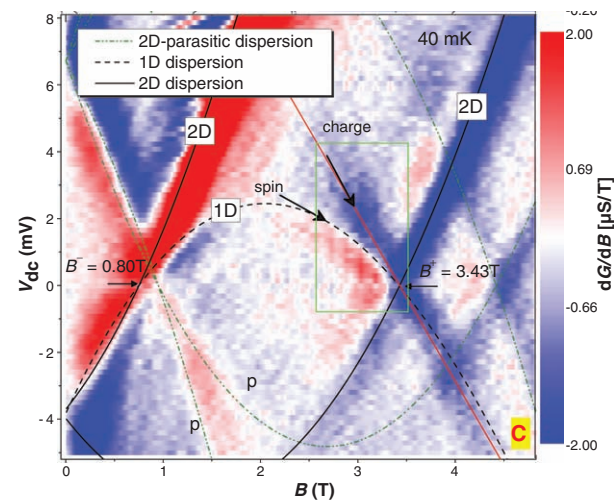
# Luttinger liquids

- One dimension: it is known that all metals are "strange"

electron spectral function develops *power-law* singularities instead of quasiparticle pole:  
many manifestations in spectroscopy, transport



Bockrath et al, nanotubes, 98



Jompol et al, GaAs quantum wires, 2009

# QM Materials

- Topological materials
- HgTe, Bi<sub>2</sub>Se<sub>3</sub>, Bi<sub>2</sub>Te<sub>3</sub>, Sb<sub>2</sub>Te<sub>3</sub>, TaAs, Cd<sub>2</sub>Se<sub>3</sub>, WTe<sub>2</sub>, MoTe<sub>2</sub>

1 H 1.00794																	2 He 4.002602				
3 Li 6.941	4 Be 9.012182															5 B 10.811	6 C 12.0107	7 N 14.00674	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797
11 Na 22.989770	12 Mg 24.3050															13 Al 26.981538	14 Si 28.0855	15 P 30.973761	16 S 32.066	17 Cl 35.4527	18 Ar 39.948
19 K 39.0983	20 Ca 40.078	21 Sc 44.955910	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938049	26 Fe 55.845	27 Co 58.933200	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92160	34 Se 78.96	35 Br 79.504	36 Kr 83.80				
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.29				
55 Cs 132.90545	56 Ba 137.327	57 La 138.9055	72 Hf 178.49	73 Ta 180.9479	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.222	78 Pt 195.078	79 Au 196.96657	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98038	84 Po (209)	85 At (210)	86 Rn (222)				
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (263)	107 Bh (262)	108 Hs (265)	109 Mt (266)	110 (269)	111 (272)	112 (277)					114 (289)	115 (287)	116 (289)	117 (289)	118 (293)	

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.035888	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

mostly s+p electron materials. Very extended, highly overlapping orbitals. Weak correlations. Heavy for strong SOC.

# QM Materials

- Topological materials

- HgTe, Bi<sub>2</sub>Se<sub>3</sub>,  
Bi<sub>2</sub>Te<sub>3</sub>, Sb<sub>2</sub>Te<sub>3</sub>, TaAs,  
Cd<sub>2</sub>Se<sub>3</sub>, WTe<sub>2</sub>,  
MoTe<sub>2</sub>

- Mn<sub>3</sub>Sn, Mn<sub>3</sub>Ge,  
YbPtBi, CeAlGe,  
Co<sub>2</sub>MnGa

<div><div>1</div><div>H</div><div>1.00794</div></div>																												<div><div>2</div><div>He</div><div>4.002602</div></div>							
<div><div>3</div><div>Li</div><div>6.941</div></div>		<div><div>4</div><div>Be</div><div>9.012182</div></div>																<div><div>5</div><div>B</div><div>10.811</div></div>		<div><div>6</div><div>C</div><div>12.0107</div></div>		<div><div>7</div><div>N</div><div>14.00674</div></div>		<div><div>8</div><div>O</div><div>15.9994</div></div>		<div><div>9</div><div>F</div><div>18.9984032</div></div>		<div><div>10</div><div>Ne</div><div>20.1797</div></div>							
<div><div>11</div><div>Na</div><div>22.989770</div></div>		<div><div>12</div><div>Mg</div><div>24.3050</div></div>																<div><div>13</div><div>Al</div><div>26.981538</div></div>		<div><div>14</div><div>Si</div><div>28.0855</div></div>		<div><div>15</div><div>P</div><div>30.973761</div></div>		<div><div>16</div><div>S</div><div>32.06</div></div>		<div><div>17</div><div>Cl</div><div>35.4527</div></div>		<div><div>18</div><div>Ar</div><div>39.948</div></div>							
<div><div>19</div><div>K</div><div>39.0983</div></div>		<div><div>20</div><div>Ca</div><div>40.078</div></div>		<div><div>21</div><div>Sc</div><div>44.955910</div></div>		<div><div>22</div><div>Ti</div><div>47.867</div></div>		<div><div>23</div><div>V</div><div>50.9415</div></div>		<div><div>24</div><div>Cr</div><div>51.9961</div></div>		<div><div>25</div><div>Mn</div><div>54.938049</div></div>		<div><div>26</div><div>Fe</div><div>55.845</div></div>		<div><div>27</div><div>Co</div><div>58.933200</div></div>		<div><div>28</div><div>Ni</div><div>58.6934</div></div>		<div><div>29</div><div>Cu</div><div>63.546</div></div>		<div><div>30</div><div>Zn</div><div>65.39</div></div>		<div><div>31</div><div>Ga</div><div>69.723</div></div>		<div><div>32</div><div>Ge</div><div>72.61</div></div>		<div><div>33</div><div>As</div><div>74.92160</div></div>		<div><div>34</div><div>Se</div><div>78.96</div></div>		<div><div>35</div><div>Br</div><div>79.904</div></div>		<div><div>36</div><div>Kr</div><div>83.80</div></div>	
<div><div>37</div><div>Rb</div><div>85.4678</div></div>		<div><div>38</div><div>Sr</div><div>87.62</div></div>		<div><div>39</div><div>Y</div><div>88.90585</div></div>		<div><div>40</div><div>Zr</div><div>91.224</div></div>		<div><div>41</div><div>Nb</div><div>92.90638</div></div>		<div><div>42</div><div>Mo</div><div>95.94</div></div>		<div><div>43</div><div>Tc</div><div>(98)</div></div>		<div><div>44</div><div>Ru</div><div>101.07</div></div>		<div><div>45</div><div>Rh</div><div>102.90550</div></div>		<div><div>46</div><div>Pd</div><div>106.42</div></div>		<div><div>47</div><div>Ag</div><div>107.8682</div></div>		<div><div>48</div><div>Cd</div><div>112.411</div></div>		<div><div>49</div><div>In</div><div>114.818</div></div>		<div><div>50</div><div>Sn</div><div>118.710</div></div>		<div><div>51</div><div>Sb</div><div>121.760</div></div>		<div><div>52</div><div>Te</div><div>127.60</div></div>		<div><div>53</div><div>I</div><div>126.90447</div></div>		<div><div>54</div><div>Xe</div><div>131.29</div></div>	
<div><div>55</div><div>Cs</div><div>132.90545</div></div>		<div><div>56</div><div>Ba</div><div>137.327</div></div>		<div><div>57</div><div>La</div><div>138.9055</div></div>		<div><div>72</div><div>Hf</div><div>178.49</div></div>		<div><div>73</div><div>Ta</div><div>180.9479</div></div>		<div><div>74</div><div>W</div><div>183.84</div></div>		<div><div>75</div><div>Re</div><div>186.207</div></div>		<div><div>76</div><div>Os</div><div>190.23</div></div>		<div><div>77</div><div>Ir</div><div>192.222</div></div>		<div><div>78</div><div>Pt</div><div>195.078</div></div>		<div><div>79</div><div>Au</div><div>196.96655</div></div>		<div><div>80</div><div>Hg</div><div>200.59</div></div>		<div><div>81</div><div>Tl</div><div>204.3833</div></div>		<div><div>82</div><div>Pb</div><div>207.2</div></div>		<div><div>83</div><div>Bi</div><div>208.98038</div></div>		<div><div>84</div><div>Po</div><div>(209)</div></div>		<div><div>85</div><div>At</div><div>(210)</div></div>		<div><div>86</div><div>Rn</div><div>(222)</div></div>	
<div><div>87</div><div>Fr</div><div>(223)</div></div>		<div><div>88</div><div>Ra</div><div>(226)</div></div>		<div><div>89</div><div>Ac</div><div>(227)</div></div>		<div><div>104</div><div>Rf</div><div>(261)</div></div>		<div><div>105</div><div>Db</div><div>(262)</div></div>		<div><div>106</div><div>Sg</div><div>(263)</div></div>		<div><div>107</div><div>Bh</div><div>(264)</div></div>		<div><div>108</div><div>Hs</div><div>(265)</div></div>		<div><div>109</div><div>Mt</div><div>(266)</div></div>		<div><div>110</div><div></div><div>(269)</div></div>		<div><div>111</div><div></div><div>(272)</div></div>		<div><div>112</div><div></div><div>(277)</div></div>						<div><div>114</div><div></div><div>(289)</div></div>		<div><div>115</div><div></div><div>(287)</div></div>		<div><div>116</div><div></div><div>(289)</div></div>		<div><div>118</div><div></div><div>(293)</div></div>	

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.03688	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

New ingredients in recent materials: partially filled 3d and 4f orbitals: correlations!

Specifically these ions host local moments

# Local moments

- Most magnetism in QMs comes from either 3d transition metal ions or 4f rare earths. These have relatively localized orbitals which don't overlap strongly with neighbors and have strong Coulomb repulsion, which localizes electrons best.

<div><div>1</div><div>H</div><div>1.00794</div></div>										<div><div>2</div><div>He</div><div>4.002602</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																											
<div><div>3</div><div>Li</div><div>6.941</div></div>		<div><div>4</div><div>Be</div><div>9.012182</div></div>												<div><div>5</div><div>B</div><div>10.811</div></div>		<div><div>6</div><div>C</div><div>12.0107</div></div>		<div><div>7</div><div>N</div><div>14.00674</div></div>		<div><div>8</div><div>O</div><div>15.9994</div></div>		<div><div>9</div><div>F</div><div>18.9984032</div></div>		<div><div>10</div><div>Ne</div><div>20.1797</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																													
<div><div>11</div><div>Na</div><div>22.989770</div></div>		<div><div>12</div><div>Mg</div><div>24.3050</div></div>												<div><div>13</div><div>Al</div><div>26.981538</div></div>		<div><div>14</div><div>Si</div><div>28.0855</div></div>		<div><div>15</div><div>P</div><div>30.973761</div></div>		<div><div>16</div><div>S</div><div>32.06</div></div>		<div><div>17</div><div>Cl</div><div>35.4527</div></div>		<div><div>18</div><div>Ar</div><div>39.948</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																													
<div><div>19</div><div>K</div><div>39.0983</div></div>		<div><div>20</div><div>Ca</div><div>40.078</div></div>		<div><div>21</div><div>Sc</div><div>44.955910</div></div>		<div><div>22</div><div>Ti</div><div>47.867</div></div>		<div><div>23</div><div>V</div><div>50.9415</div></div>		<div><div>24</div><div>Cr</div><div>51.9961</div></div>		<div><div>25</div><div>Mn</div><div>54.938049</div></div>		<div><div>26</div><div>Fe</div><div>55.845</div></div>		<div><div>27</div><div>Co</div><div>58.933200</div></div>		<div><div>28</div><div>Ni</div><div>58.6934</div></div>		<div><div>29</div><div>Cu</div><div>63.546</div></div>		<div><div>30</div><div>Zn</div><div>65.39</div></div>		<div><div>31</div><div>Ga</div><div>69.723</div></div>		<div><div>32</div><div>Ge</div><div>72.61</div></div>		<div><div>33</div><div>As</div><div>74.92160</div></div>		<div><div>34</div><div>Se</div><div>78.96</div></div>		<div><div>35</div><div>Br</div><div>79.504</div></div>		<div><div>36</div><div>Kr</div><div>83.80</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
<div><div>37</div><div>Rb</div><div>85.4678</div></div>		<div><div>38</div><div>Sr</div><div>87.62</div></div>		<div><div>39</div><div>Y</div><div>88.90585</div></div>		<div><div>40</div><div>Zr</div><div>91.224</div></div>		<div><div>41</div><div>Nb</div><div>92.90638</div></div>		<div><div>42</div><div>Mo</div><div>95.94</div></div>		<div><div>43</div><div>Tc</div><div>(98)</div></div>		<div><div>44</div><div>Ru</div><div>101.07</div></div>		<div><div>45</div><div>Rh</div><div>102.90550</div></div>		<div><div>46</div><div>Pd</div><div>106.42</div></div>		<div><div>47</div><div>Ag</div><div>107.8682</div></div>		<div><div>48</div><div>Cd</div><div>112.411</div></div>		<div><div>49</div><div>In</div><div>114.818</div></div>		<div><div>50</div><div>Sn</div><div>118.710</div></div>		<div><div>51</div><div>Sb</div><div>121.760</div></div>		<div><div>52</div><div>Te</div><div>127.60</div></div>		<div><div>53</div><div>I</div><div>126.90447</div></div>		<div><div>54</div><div>Xe</div><div>131.29</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
<div><div>55</div><div>Cs</div><div>132.90545</div></div>		<div><div>56</div><div>Ba</div><div>137.327</div></div>		<div><div>57</div><div>La</div><div>138.9055</div></div>		<div><div>72</div><div>Hf</div><div>178.49</div></div>		<div><div>73</div><div>Ta</div><div>180.9479</div></div>		<div><div>74</div><div>W</div><div>183.84</div></div>		<div><div>75</div><div>Re</div><div>186.207</div></div>		<div><div>76</div><div>Os</div><div>190.23</div></div>		<div><div>77</div><div>Ir</div><div>192.217</div></div>		<div><div>78</div><div>Pt</div><div>195.078</div></div>		<div><div>79</div><div>Au</div><div>196.96655</div></div>		<div><div>80</div><div>Hg</div><div>200.59</div></div>		<div><div>81</div><div>Tl</div><div>204.3833</div></div>		<div><div>82</div><div>Pb</div><div>207.2</div></div>		<div><div>83</div><div>Bi</div><div>208.98038</div></div>		<div><div>84</div><div>Po</div><div>(209)</div></div>		<div><div>85</div><div>At</div><div>(210)</div></div>		<div><div>86</div><div>Rn</div><div>(222)</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																			
<div><div>87</div><div>Fr</div><div>(223)</div></div>		<div><div>88</div><div>Ra</div><div>(226)</div></div>		<div><div>89</div><div>Ac</div><div>(227)</div></div>		<div><div>104</div><div>Rf</div><div>(261)</div></div>		<div><div>105</div><div>Db</div><div>(262)</div></div>		<div><div>106</div><div>Sg</div><div>(263)</div></div>		<div><div>107</div><div>Bh</div><div>(262)</div></div>		<div><div>108</div><div>Hs</div><div>(265)</div></div>		<div><div>109</div><div>Mt</div><div>(266)</div></div>		<div><div>110</div><div></div><div>(269)</div></div>		<div><div>111</div><div></div><div>(272)</div></div>		<div><div>112</div><div></div><div>(277)</div></div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																															

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.035888	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

# Local moments

- In 3d transition metals, *usually* magnetism is fairly isotropic, i.e. spins are “Heisenberg like”, because crystal fields split the d orbitals and spin-orbit coupling is relatively weak (Co is most common exception, when very localized). Exchange interactions between spins vary from quite strong (1000K) to quite weak (1K).

1 H 1.00794																	2 He 4.002602			
3 Li 6.941	4 Be 9.012182											5 B 10.811	6 C 12.0107	7 N 14.00674	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797			
11 Na 22.989770	12 Mg 24.3050											13 Al 26.981538	14 Si 28.0855	15 P 30.973761	16 S 32.066	17 Cl 35.4527	18 Ar 39.948			
19 K 39.0983	20 Ca 40.078	21 Sc 44.955910	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938049	26 Fe 55.845	27 Co 58.933200	28 Ni 58.6934	29 Cu 63.545	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92160	34 Se 78.96	35 Br 79.504	36 Kr 83.80			
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.29			
55 Cs 132.90545	56 Ba 137.327	57 La 138.9055	72 Hf 178.49	73 Ta 180.9479	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.217	78 Pt 195.078	79 Au 196.96655	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98038	84 Po (209)	85 At (210)	86 Rn (222)			
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (263)	107 Bh (262)	108 Hs (265)	109 Mt (266)	110 (269)	111 (272)	112 (277)					114 (289)	115 (287)	116 (289)	117 (289)	118 (293)

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.035888	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

# Local moments

- In 4f lanthanides, spin-orbit coupling is dominant over crystal fields and so magnetic moments become large (incorporating orbital moment) and often very anisotropic (due to large SOC). They have complex multiplet structures, and weak exchange interactions.

1 H 1.00794																	2 He 4.002602
3 Li 6.941	4 Be 9.012182											5 B 10.811	6 C 12.0107	7 N 14.00674	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797
11 Na 22.989770	12 Mg 24.3050											13 Al 26.981538	14 Si 28.0855	15 P 30.973761	16 S 32.066	17 Cl 35.4527	18 Ar 39.948
19 K 39.0983	20 Ca 40.078	21 Sc 44.955910	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938049	26 Fe 55.845	27 Co 58.933200	28 Ni 58.6934	29 Cu 63.545	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92160	34 Se 78.96	35 Br 79.504	36 Kr 83.80
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.29
55 Cs 132.90545	56 Ba 137.327	57 La 138.9055	72 Hf 178.49	73 Ta 180.9479	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.217	78 Pt 195.078	79 Au 196.96655	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98038	84 Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (263)	107 Bh (262)	108 Hs (265)	109 Mt (266)	110 (269)	111 (272)	112 (277)	(289)		(289)		(293)	

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.035888	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

# QM Materials

- Quantum spin liquids and interesting insulating antiferromagnets

<div>1 H 1.00794</div>										<div>2 He 4.002602</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																									
<div>3 Li 6.941</div>		<div>4 Be 9.012182</div>												<div>5 B 10.811</div>		<div>6 C 12.0107</div>		<div>7 N 14.00674</div>		<div>8 O 15.9994</div>		<div>9 F 18.9984032</div>		<div>10 Ne 20.1797</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																											
<div>11 Na 22.989770</div>		<div>12 Mg 24.3050</div>												<div>13 Al 26.981538</div>		<div>14 Si 28.0855</div>		<div>15 P 30.973761</div>		<div>16 S 32.06</div>		<div>17 Cl 35.4527</div>		<div>18 Ar 39.948</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																											
<div>19 K 39.0983</div>		<div>20 Ca 40.078</div>		<div>21 Sc 44.955910</div>		<div>22 Ti 47.867</div>		<div>23 V 50.9415</div>		<div>24 Cr 51.9961</div>		<div>25 Mn 54.938049</div>		<div>26 Fe 55.845</div>		<div>27 Co 58.933200</div>		<div>28 Ni 58.6934</div>		<div>29 Cu 63.546</div>		<div>30 Zn 65.39</div>		<div>31 Ga 69.723</div>		<div>32 Ge 72.61</div>		<div>33 As 74.92160</div>		<div>34 Se 78.96</div>		<div>35 Br 79.504</div>		<div>36 Kr 83.80</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
<div>37 Rb 85.4678</div>		<div>38 Sr 87.62</div>		<div>39 Y 88.90585</div>		<div>40 Zr 91.224</div>		<div>41 Nb 92.90638</div>		<div>42 Mo 95.94</div>		<div>43 Tc (98)</div>		<div>44 Ru 101.07</div>		<div>45 Rh 102.90550</div>		<div>46 Pd 106.42</div>		<div>47 Ag 107.8652</div>		<div>48 Cd 112.411</div>		<div>49 In 114.818</div>		<div>50 Sn 118.710</div>		<div>51 Sb 121.760</div>		<div>52 Te 127.60</div>		<div>53 I 126.90447</div>		<div>54 Xe 131.29</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
<div>55 Cs 132.90545</div>		<div>56 Ba 137.327</div>		<div>57 La 138.9055</div>		<div>72 Hf 178.49</div>		<div>73 Ta 180.9479</div>		<div>74 W 183.84</div>		<div>75 Re 186.207</div>		<div>76 Os 190.23</div>		<div>77 Ir 192.217</div>		<div>78 Pt 195.078</div>		<div>79 Au 196.96655</div>		<div>80 Hg 200.59</div>		<div>81 Tl 204.3833</div>		<div>82 Pb 207.2</div>		<div>83 Bi 208.98038</div>		<div>84 Po (209)</div>		<div>85 At (210)</div>		<div>86 Rn (222)</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																	
<div>87 Fr (223)</div>		<div>88 Ra (226)</div>		<div>89 Ac (227)</div>		<div>104 Rf (261)</div>		<div>105 Db (262)</div>		<div>106 Sg (263)</div>		<div>107 Bh (262)</div>		<div>108 Hs (265)</div>		<div>109 Mt (266)</div>		<div>110 (269)</div>		<div>111 (272)</div>		<div>112 (277)</div>																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																																													

58 Ce 140.116	59 Pr 140.90768	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.036888	92 U 238.0289	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

- $\text{ZnCu}_3(\text{OH})_6\text{Cl}_2$ , a-  
 $\text{RuCl}_3$ ,  $\text{Pr}_2\text{Zr}_2\text{O}_7$ ,  
 $\text{Cs}_2\text{CuCl}_4$ ,  $\text{Yb}_2\text{Ti}_2\text{O}_7$

# QM Materials

- Correlated metals/  
Mott transitions
- $\text{RTiO}_3, \text{RVO}_3, \text{RNiO}_3,$   
 $\text{RMnO}_3, \text{RCoO}_3, \dots$
- $\text{Cd}_2\text{Os}_2\text{O}_7, \text{Nd}_2\text{Ir}_2\text{O}_7,$   
 $\text{Sr}_2\text{IrO}_4, \text{Sr}_2\text{RuO}_4, \dots$
- $\text{URu}_2\text{Si}_2, \text{CeAl}_3, \dots$

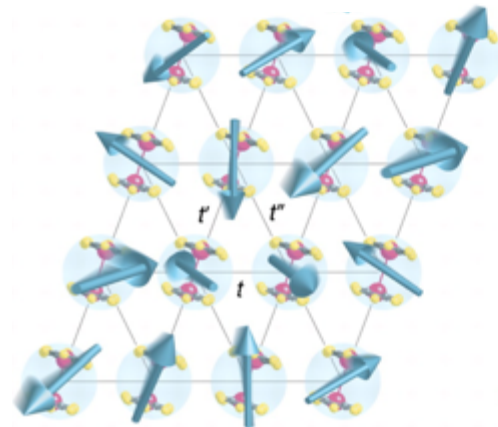
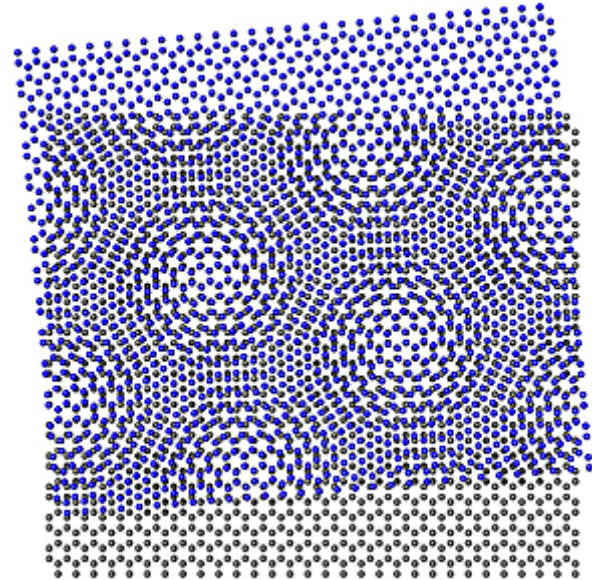
1 H 1.00794																	2 He 4.002602				
3 Li 6.941	4 Be 9.012182															5 B 10.811	6 C 12.0107	7 N 14.00674	8 O 15.9994	9 F 18.9984032	10 Ne 20.1797
11 Na 22.989770	12 Mg 24.3050															13 Al 26.981538	14 Si 28.0855	15 P 30.973761	16 S 32.066	17 Cl 35.4527	18 Ar 39.948
19 K 39.0983	20 Ca 40.078	21 Sc 44.955910	22 Ti 47.867	23 V 50.9415	24 Cr 51.9961	25 Mn 54.938049	26 Fe 55.845	27 Co 58.933200	28 Ni 58.6934	29 Cu 63.546	30 Zn 65.39	31 Ga 69.723	32 Ge 72.61	33 As 74.92160	34 Se 78.96	35 Br 79.504	36 Kr 83.80				
37 Rb 85.4678	38 Sr 87.62	39 Y 88.90585	40 Zr 91.224	41 Nb 92.90638	42 Mo 95.94	43 Tc (98)	44 Ru 101.07	45 Rh 102.90550	46 Pd 106.42	47 Ag 107.8682	48 Cd 112.411	49 In 114.818	50 Sn 118.710	51 Sb 121.760	52 Te 127.60	53 I 126.90447	54 Xe 131.29				
55 Cs 132.90545	56 Ba 137.327	57 La 138.9055	72 Hf 178.49	73 Ta 180.9479	74 W 183.84	75 Re 186.207	76 Os 190.23	77 Ir 192.217	78 Pt 195.078	79 Au 196.96655	80 Hg 200.59	81 Tl 204.3833	82 Pb 207.2	83 Bi 208.98038	84 Po (209)	85 At (210)	86 Rn (222)				
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (263)	107 Bh (262)	108 Hs (265)	109 Mt (266)	110 (269)	111 (272)	112 (277)					114 (289)	115 (289)	116 (289)	117 (289)	118 (293)	

58 Ce 140.116	59 Pr 140.90765	60 Nd 144.24	61 Pm (145)	62 Sm 150.36	63 Eu 151.964	64 Gd 157.25	65 Tb 158.92534	66 Dy 162.50	67 Ho 164.93032	68 Er 167.26	69 Tm 168.93421	70 Yb 173.04	71 Lu 174.967
90 Th 232.0381	91 Pa 231.03688	92 U 238.02891	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

transition metals and  
rare earths mostly.

# QM Materials

- Twisted graphene, organics “break the mold”
- Become correlated because large unit cell suppresses hopping/bandwidth
- “designed” QMs



Thanks